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Shape memory effect and superelastic behavior of TiNi shape memory alloy processed by vacuum plasma spray method

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12 Abstract

Machining of TiNi shape memory alloy into a complicated three-dimensional (3D) shape is quite difficult, thus a near-net shape forming of 3D shaped TiNi alloy is attractive and cost-effective. Vacuum plasma spray (VPS) process is one of such near-net shape forming processes. In this paper, two kinds of thick TiNi layer, Ni-rich and Ti-rich compositions, were fabricated by VPS process and their shape memory effect (SME) and superelastic (SE) behaviors were characterized. As-VPS processed Ni-rich TiNi which was subjected to homogenization at 1163 K for 7.2 ks and subsequent aging at 773 K for 18 ks exhibited good martensitic transformation behavior. The recoverable strain of the TiNi alloys due to SME and SE behavior were measured to be 2.4% and 5.0%, respectively. The Ti-rich TiNi alloy also exhibited good martensitic transformation behavior and SME in the as-homogenized state.

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21 Keywords: Vacuum plasma spray (VPS); Near-net shape; Stress-strain curve; Superelastic behavior; Shape memory effect; Martensitic transformation; TiNi alloy

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1. Introduction

TiNi shape memory alloys are widely used in many appli-2 cations such actuators, due to their excellent shape memory 3 4 effect (SME) and superelastic (SE) behaviors. However, the TiNi alloys have poor cold workability due to their high work-5 hardening rate [1]. An elongation of Ti-50.0 at.% Ni alloy is 6 about 18% at room temperature. When Ni content exceeds 7 51.0 at.%, the cold working of such TiNi alloy becomes very 8 difficult [2]. Therefore, hot working at around 1073 K is nor-9 mally applied to TiNi alloys, which inevitably gives rise to thick 10 oxide layer on the TiNi surface, degrading their SME and SE 11 properties. Due to the difficulty of the machining of TiNi alloys, 12 we must use a tungsten carbide tool whose use life is short, thus, 13 often we end up with using electron discharge machining which 14 is more costly. One of the solutions to overcome these problems 15 is use of near-net shape process by powder metallurgy (PM). 16

Several researchers reported SME and SE behaviors of as-PM processed TiNi alloys [3–7]. Kato et al. reported about 6% SE and 7% SME strains in TiNi alloy that was hot-isotropically pressed [3,4]. It is noted that the material produced by PM tends to eliminate internal voids. In addition, the sintering usually utilizes solid–solid reaction between powders with the results that the particle boundary is weaker than grain boundary, leading to reduction in the mechanical properties of as-sintered materials. Particularly, fracture strain and fatigue properties of PM-processed materials are reduced [4,5].

Vacuum plasma spray (VPS) process is one of the spray-27 ing methods. Source powders are injected to plasma stream 28 where the injected material is melted immediately by a high 29 temperature plasma jet, and then the molten metal particles are 30 accelerated in the plasma jet stream to deposit as overlapping 31 "splats" onto a substrate to form a P/M consolidated deposit. The 32 source powders are melted by the plasma jets and the sprayed 33 layers become high density. However, during VPS process oxide 34 and nitrides more likely to form, due to the extremely high tem-35 perature plasma jet. In particular, titanium in a TiNi alloy has a 36 strong affinity with oxygen, so that titanium oxides or nitrides 37

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Table 1

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Results of composition analysis of source powders and sprayed TiNi alloy				
	NiTi sample (Ni-rich)		TiNi sample (Ti-rich)	
	Original TiNi powder (wt.%)	Sprayed TiNi (wt.%)	Original TiNi powder (wt.%)	Sprayed TiNi (wt
Ni	56.8 (51.9 at.%)	55.6 (51.3 at.%)	54.7 (49.7 at.%)	54.4 (49.5 at.%)
Ti	43.0 (48.1 at.%)	42.9 (48.7 at.%)	45.2 (50.3 at.%)	45.2 (50.5 at.%)
0	0.120	0.230	0.066	0.101
Ν	0.002	0.006	0.003	0.025
С	0.065	0.11	0.011	0.008
Others	Balance	Balance	Balance	Balance

can be formed easily during the process. Thus, control of atmo-38 sphere, preferably in high vacuum is desired. Characterization 39 of as-VPS processed TiNi alloys has been reported by several 40 researchers [8–11]. However, the previous studies were focused 41 on very thin film TiNi (under $100 \,\mu m$), where the formation of 42 oxide, nitride and other intermetallic compounds were observed. 43 There have been no studies on SME and SE behavior of sprayed 44 TiNi alloys of relatively large size. 45

In this paper, the experimental results of VPS processing of 46 TiNi alloys under low oxygen atmosphere will be reported and 47 their heat treatment effects on SME and SE behaviors assessed 48 by compression testing will be examined. The microstructures 49 of as-VPS processed, heat-treated TiNi alloys are characterized 50 by XRD and TEM with aim of elucidating the merits of VPS 51 processing of TiNi alloys and the mechanisms responsible for 52 their improved SME and SE properties. 53

2. Experimental procedure

Two kinds of TiNi alloy powders, Ni-rich composition and 55 Ti-rich composition, were sprayed on a steel pipe substrate of 56 outer diameter being 31.8 mm with 1.7 mm wall thickness. The 57 sprayed TiNi layers were more than 15 mm thick. The compo-58 sition of powders and sprayed materials are listed in Table 1. 59 Hereafter, the Ni-rich sample and Ti-rich sample are called as 60 'NiTi' and 'TiNi', respectively. The composition of the sprayed 61 materials is found to be the same as those of the starting pow-62 ders. Contaminations by oxygen, nitrogen and carbon were also 63 observed in the powders and sprayed materials. However, the 64 level of contaminations of NiTi and TiNi remained the same 65 as the source powders. For characterization of sprayed materi-66 als, the surface layer of as-VPS processed material was taken 67 out from substrate and machined to cylindrical bar of diameter 68 5 mm. The machined cylinders were subjected to homogenized 69 heat treatment at 1163 K for 7.2 ks under vacuum atmosphere. 70 Subsequent aging treatments were also carried out to some of the 71 homogenized materials in air or vacuum, followed by water or 72 helium quenching. Transformation temperatures of the as-VPS 73 processed material with and without heat treatment were mea-74 sured by differential scanning calorimetry (Perkin-Elmer, DSC 75 6) with heating and cooling rate of 20 K/min and 10 K/min, 76 respectively. The measured transformation temperatures were 77 martensitic transformation start (M_s) and finish (M_f) , austenitic 78 transformation start (A_s) and finish (A_f) and R-phase trans-79 formation start (R_s) and finish (R_f) . After the heat treatment, 80

compression testing, limited thermal cycle testing and stress 81 cycle testing were applied to selected specimens. The speci-82 mens for the mechanical testing were machined to cylindrical 83 shape with 5 mm in diameter and 5 mm in height using elec-84 trical discharging machine. X-ray diffraction (XRD) was also 85 performed to determine the formed phases in the VPS processed 86 specimens using Cu Ka radiation. Microstructural observations 87 were conducted on selected specimens by transmission elec-88 tron microscope (TEM) operated at 120 kV. Samples for TEM 89 observations were prepared by electro-chemical polishing using 90 H_2SO_4 :methanol = 2:8 (vol.) solution. 91

TiNi (wt.%)

92

3. Experimental results

The XRD results of as-sprayed and as-homogenized NiTi 93 and TiNi are shown in Fig. 1. The as-sprayed NiTi is found to 94 be predominantly composed of austenite phase. In addition, the 95 minor peaks, indicated by open squares in Fig. 1, correspond-96 ing to martensite phase are seen in the vicinity of the austenite 97 phase peak. This is attributed to non-uniform distribution of 98 TiNi composition. After the homogenization, the XRD pattern 99 is found to be similar to the as-sprayed state, suggesting that 100 some degree of the unevenness of composition still remains after 101 the homogenization. In the case of TiNi specimen, its majority 102 is composed of martensite phase with austenite phase as minor-103 ity. Even though the homogenization treatment was applied, the 104 TiNi specimen is still composed of martensite phase and austen-105 ite phase. Therefore, the TiNi specimen still has an unevenness of 106 composition after the homogenization. However, in both NiTi 107 and TiNi samples, no unfavorable oxides, nitrides and inter-108 metallic compound phases were observed in theses XRD results. 109 These results indicate that the VPS processed materials are com-110 posed by TiNi austenite phase and martensite phase dominantly, 111 and the amounts of other unfavorable phases remain very mod-112 est. Therefore, this demonstrates that VPS process provides 113 clean NiTi and TiNi materials. 114

The DSC curves of as-sprayed and as-homogenized NiTi 115 and TiNi are shown in Fig. 2. In the case of NiTi, no sig-116 nificant transformation peaks were observed in both of the 117 as-sprayed and as-homogenized specimens. This is because 118 the transformation temperatures were lower than the temper-119 ature range measured by DSC. The estimated M_s temperature 120 from the composition of NiTi specimen was below 170 K [12]. 121 In contrast, TiNi in the as-sprayed state exhibits the marten-122 sitic and reverse martensitic transformation behavior, where 123

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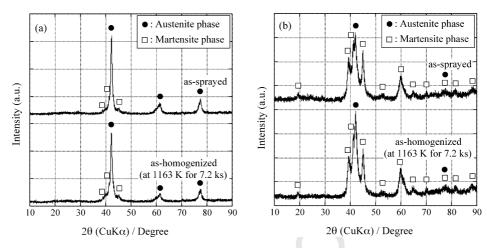


Fig. 1. XRD profiles of as-sprayed and as-homogenized state in (a) NiTi sample and (b) TiNi sample.

the measured transformation temperatures were $M_s = 339$ K, 124 $M_{\rm f}$ = 320 K, $A_{\rm s}$ = 351 K and $A_{\rm f}$ = 379 K. After the homogeniza-125 tion treatment, the transformation peaks become sharper than 126 that of as-sprayed material, and all of transformation temper-127 atures were increased as compared with the as-VPS processed 128 specimen. The transformation temperatures of as-homogenized 129 TiNi specimen were $M_s = 342$ K, $M_f = 330$ K, $A_s = 363$ K and 130 $A_{\rm f}$ = 385 K. The changes in the DSC profiles implied that the 131 unevenness of composition was alleviated by the homoge-132 nization. The enthalpy changes in reverse transformation in 133 as-sprayed and as-homogenized TiNi materials are 26.10 and 134 25.50 J/g, respectively. These values are found to be higher than 135 136 those of bulk TiNi alloy previously reported [13].

Fig. 3 shows the changes in the transformation temperatures by the aging at 773 K in the NiTi specimen. The as-homogenized sample (Fig. 3(a)) did not show any peaks. The heat treatment at 773 K for 2.4 ks in air (Fig. 3(b)) caused the R-phase transition and martensite phase transition on the cooling, and austenite phase transition on the heating. However, the transformation peaks are broad. By increasing aging time (7.2 ks, Fig. 3(c)), 143 the shapes of all peaks become sharper, and the transformation 144 temperatures are increased. After 18 ks aging (Fig. 3(d)), the 145 shapes of peaks are slightly sharper than those of the specimen 146 with 7.2 ks aging. However, all of transformation temperatures 147 except for $M_{\rm f}$ are almost the same as those of the 7.2 ks aged 148 sample. The transformation temperatures after aging for 18 ks 149 are $M_s = 262 \text{ K}, M_f = 249 \text{ K}, A_s = 294 \text{ K}, A_f = 311, R_s = 320 \text{ K}$ and 150 $R_{\rm f}$ = 293 K. Fig. 3(e) shows the DSC behavior of the samples 151 aged in a vacuum atmosphere at 773 K for 18 ks followed by He 152 quenching. The transformation behavior is similar to the aged 153 sample under air. However, the transformation temperatures of 154 (e) are higher than those of (d). The reason for change in transfor-155 mation temperature by aging in the NiTi specimen is considered 156 to be due to the crystallization of amorphous phase or precipita-157 tion. In this case, we believe that the change in the transformation 158 temperatures were due to precipitation, because the Ti₃Ni₄ pre-159 cipitates were observed by TEM (see Fig. 4), and no amorphous 160 phase was detected by XRD and TEM. The detailed mechanism 161

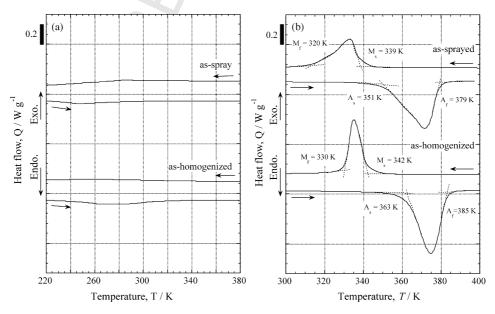


Fig. 2. DSC curves of as-sprayed and as-homogenized state in (a) NiTi sample and (b) TiNi sample.

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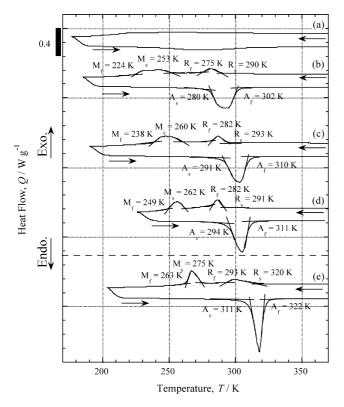


Fig. 3. DSC curves of NiTi sample after (a) homogenization, and subsequent aging at 773 K (b) for 2.4 ks, (c) for 7.2 ks, (d) for 18 ks under air and (e) for 18 ks under vacuum condition.

of the change in transformation temperatures by precipitationsis described later.

Fig. 4 shows the TEM bright field image of the NiTi sample which was homogenized and aged at 773 K for 18 ks under air, where the selected area diffraction (SAD) pattern obtained from the image is show the right. The electron beam is almost parallel to $[1 1 0]_{BZ}$ direction. The several precipitates with lenticular shape are seen. These precipitates were not observed in the sample of as-homogenized state. By the analysis of the SAD 170 pattern, it is confirmed that the lenticular precipitates exist along 171 $(1 \ 1 \ 1)_{B2}$ plane. The metastable Ti₃Ni₄ phase is known to have 172 lenticular shape and they are oriented along the $(1 \ 1 \ 1)_{B2}$ due 173 to their coherency with B2 matrix [14]. Therefore, the precipi-174 tates observed by this TEM study are identified as Ti₃Ni₄. The 175 formation of Ni-rich precipitation can lead to increasing of Ti 176 content in TiNi matrix. Martensitic and reverse transformation 177 temperatures are sensitive to alloy composition. Reduction of 178 0.1 at.% Ni leads to an increment of transformation temperature 179 of about 10 K [12]. Therefore, the transformation temperature 180 of the NiTi sample is increased by a change in the composition 181 of TiNi matrix due to aging effect. The enthalpy of transfor-182 mations of Fig. 3(d) obtained from the area of DSC peaks are 183 -8.49 J/g for martensitic transformation, -4.12 J/g for R-phase 184 transformation and 14.64 J/g for reverse transformation. The val-185 ues of the enthalpy of the reverse transformation (14.64 J/g) is 186 comparable to that reported on reverse transformation in a bulk 187 TiNi alloy (15.80 J/g) [13], indicating that the sprayed TiNi alloy 188 exhibits SME and SE properties similar to bulk materials. The 189 effects of the aging treatment were also examined for TiNi spec-190 imen. No significant change in transformation temperatures was 19 found in the aged TiNi sample. The similar results were reported 192 [15]. This is, probably, due to the difference in the precipitation 193 process between TiNi and NiTi. 194

The results of compression testing of NiTi aged at 773 K for 195 18 ks under air are shown in Fig. 5, where (a) is the stress-strain 196 curve tested at $M_{\rm f}$ temperature (249 K) which exhibits SME 197 behavior. The residual strain after un-loading (2.4%) becomes 198 zero upon temperature increase above $A_{\rm f}$ temperature, as shown 199 by arrow, indicating perfect SME behavior, Fig. 5(a). The 200 stress-strain curve tested near Af temperature (304 K) shown in 201 Fig. 5(b) exhibits good SE behavior. The onset stress and strain 202 of the stress-induced martensite are 400 MPa and 2.2%, respec-203 tively. The strain of slightly over 5.0% can be recovered after 204 un-loading, which is a large superelastic strain. By increasing 205

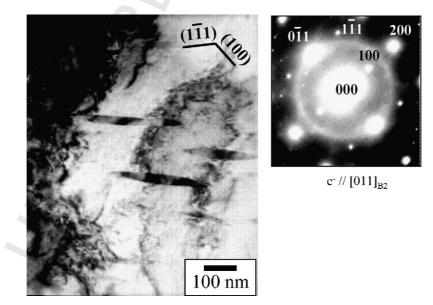


Fig. 4. TEM image of NiTi sample aging at 773 K for 18 ks under air, where incident electron beam is almost parallel to [011]_{B2} direction.

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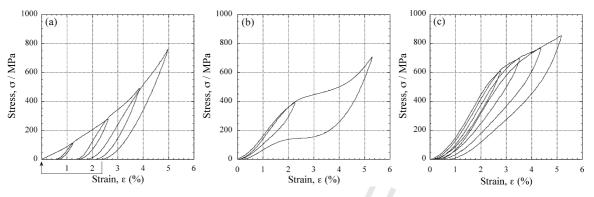


Fig. 5. Stress-strain curves of NiTi sample aged at 773 K for 18 ks under air tested at (a) 249 K, (b) 304 K and (c) 320 K.

testing temperature to above $A_{\rm f}$ (320 K) (Fig. 5(c)), the superelastic behavior is still observed, but with unrecoverable strain of 0.8%. This is because the increased testing temperature leads to increase the onset stress for the stress-induced martensite, while reducing the yield stress, hence resulting in partial plastic deformation.

The stress-strain curves of TiNi in as-homogenized state are 212 shown in Fig. 6. The TiNi sample tested at room temperature 213 (RT) exhibits SME effect. The recoverable SME strain by heat-214 ing at 473 K after deformation is about 4.0% as indicated by 215 arrow. This value is comparable to that of a bulk TiNi shape 216 memory alloy. It is noted in Fig. 6(b) that TiNi sample tested at 217 around A_f temperature (384 K) does not show any SE behavior, 218 i.e. only plastic deformation is observed. This is attributed to the 219 fact that the high testing temperature decreases the yield strength 220 of the TiNi austenite phase, which is below the onset stress for 22 stress-induced martensite, resulting in the plastic deformation 222 before the formation of stress-induced martensite phase. Liu et 223 al. reported the plastic yielding of TiNi alloy above 350 K, whose 224 $M_{\rm s}$ temperature was about 305 K, due to the same mechanism 225 mentioned above [16]. 226

Fig. 7(a) shows the result of thermal cyclic testing of NiTi sample aged at 773 K for 18 ks under vacuum. The pre-stress compression of 250 MPa was applied on NiTi sample at RT and kept during the test. From the stress–strain curve of the sample shown in Fig. 7(b), it is clear that the sample was in stress-induced martensite state under pre-stress as indicated 232 by solid triangle. The strain recovers due to reverse trans-233 formation at 330 K on first heating and deformed again by 234 martensitic transformation at 297 K on cooling. These testing 235 temperatures are higher than the A_s and M_s detected by the 236 DSC without pre-stress, see Fig. 3(e). The constant stress can 237 promote the martensitic and reverse martensitic transformation 238 that take place at higher temperatures. The change in transfor-239 mation temperature under constant stress can be estimated by 240 Clausisu–Clapeyron equation as follows [17]: 241

$$\frac{\mathrm{d}\sigma}{\mathrm{d}T} = -\frac{\Delta H}{\varepsilon T},\tag{1}$$

where $\Delta H (= -12.45 \text{ J/g})$ is an enthalpy change due to marten-243 sitic transformation, ε ($\approx 2.5\%$) a transformation strain and T 244 $(\approx (M_{\rm s} + A_{\rm s})/2)$ is an equilibrium temperature. The estimated 245 value of $d\sigma/dT$ based on $\Delta H = -12.45$ J/g is 11.0 MPa/K, which 246 is in good agreement with the experimental value obtained from 247 Fig. 7(a) $(d\sigma/dT = 250 \text{ MPa}/22 \text{ K} = 11.4 \text{ MPa}/\text{K})$. The recovered 248 strain by first heating is about 1.2%, Fig. 7(a). However, on 249 the cooling, the strain does not revert to zero, i.e. about 0.13%250 permanent strain remained in the sample at the end of first 251 cooling cycle. By increasing thermal cycles under constant 252 stress, the permanent strain increases. After 15 cycles, the 253 strain-temperature curve is almost saturated. The recoverable 254 strain after 15 cycles is about 0.6%. Even though the amount 255

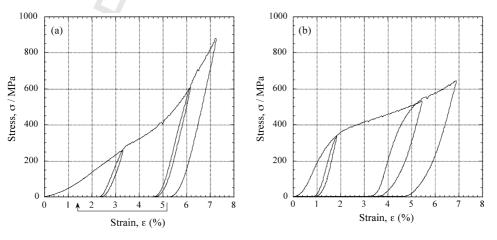


Fig. 6. Stress-strain curves of TiNi sample in as-homogenized state tested at (a) RT and (b) 384 K.

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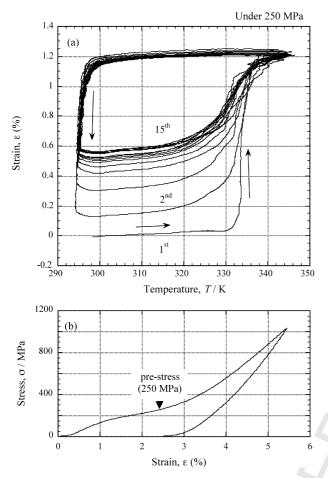


Fig. 7. (a) Results of thermal cyclic testing under constant stress in aged NiTi sample at 773 K for 18 ks under vacuum. (b) Stress–strain curve of the NiTi sample where the state of sample under pre-stress compression is indicated by solid triangle.

of the recoverable strain is as small as 0.6%, the repeatable
strain-temperature cycle behavior after 15 cycles is considered
useful for use in some applications.

The results of thermal cyclic testing of TiNi sample aged at 259 773 K for 18 ks under vacuum condition are shown in Fig. 8. The 260 TiNi sample was pre-deformed by 250 MPa compression stress 261 at RT, where the sample was deformed in martensite state shown 262 in Fig. 8(b). On the first heating, the sample recovers at 413 K. 263 However, the SME strain is small, which is about 0.6%, Fig. 8(a). 264 On the first cooling, the TiNi sample starts to shrink at 333 K 265 due to martensitic transformation. By furt cooling, the sam-266 ple shrinks to shorter length than the iNima length. After the 267 second cycle, the temperature-strain curves exhibit hysteresis 268 loop. The permanent strain is accumulated by increasing cycle 269 number, while recoverable strain is decreased with increasing 270 cycle number. The hysteresis loop is almost saturated after 15th 271 cycle. The available strain after 15 cycles is about 2.2%, which 272 is relatively large, thus useful range for several applications. 273 The small SME strain on the first heating is due to the plastic 274 deformation of the sample. The schematic relationship between 275 the reverse transformation temperatures $(A_s \text{ and } A_f)$ and yield 276 stress (σ_v) is shown in Fig. 9. In the case of TiNi sample hav-277 ing relatively higher transformation temperatures, the stress at 278

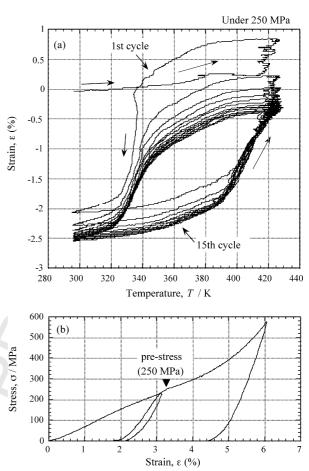
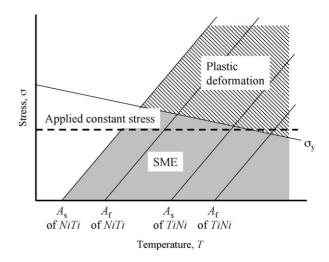
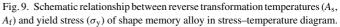


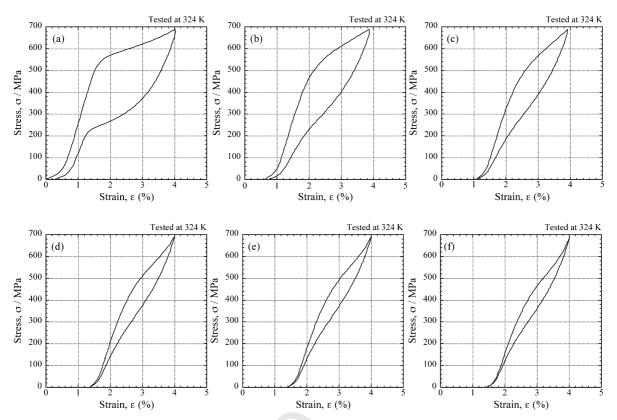
Fig. 8. (a) Results of thermal cyclic testing performed under constant stress in TiNi sample where the samples are homogenized and aged at 773 K for 18 ks under vacuum before testing. (b) Stress–strain curve of the sample tested at RT where the state of sample under pre-stress compression is indicated by solid triangle.

 $A_{\rm s}$ temperature and yield stress are nearly the same under the applied constant stress condition, thus the sample can go into plastic deformation (hatched) region by increasing temperature before achieving the stress at $A_{\rm f}$. Here, the direction of SME





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Fig. 10. Changes in stress-strain curves by stress cycles in NiTi sample aged at 773 K for 18 ks under vacuum after (a) 1st, (b) 2nd, (c) 10th, (d) 50th, (e) 100th and (f) 200th cycles.

strain is positive, and that of plastic deformation is negative. Therefore, the part of SME strain is canceled out by plastic deformation of the austenite phase. On the other hand, the NiTi sample which has lower transformation temperatures, so that the points of A_s and A_f can exist in the SME (gray) region and larger SME strain was observed in the first heating (see Fig. 7(a)).

The results of stress cycle testing of the NiTi sample aged 289 at 773 K for 18 ks under vacuum are shown in Fig. 10. At the 290 first cycle, the sample exhibits SE behavior with the onset stress 291 of stress-induced martensite being 540 MPa. The first plateau 292 with less steep slope at the beginning of stress-strain curve 293 with 0-0.4% strain may be due to the stress-induced R-phase 294 transformation, because the testing temperature (324 K) is very 295 close to R_s temperature (see Fig. 3(e)). After the unloading, 296 about 0.3% permanent strain is observed. By increasing number 297 of cycles, the permanent strain is accumulated up to 1.4% by 298 50 cycles. The hysteresis loop in the stress-strain curves after 290 50 cycles becomes narrower, and the plateau due to R-phase 300 disappeared and the stress-induced martensite transformation 301 plateau being steeper. The sample after 200 cycles still exhibits 302 2.6% recoverable strain. Furthermore, no cracking was observed 303 on the sample surface after 200 stress cycles. The onset stress 304 of stress-induced martensite decreases with increasing number 305 of cycle, i.e. about 540 MPa at the first cycle being reduced to 306 about 380 MPa after 200 cycles. The strain of plateau region also 307 decreases with increasing number of cycle. By uni-axial cyclic 308 testing, certain dislocations, which provide favorable stress field 309 for formation of martensite, were introduced in the sample, so 310 as to decrease the onset stress for stress-induced martensite.

4. Conclusion

The thick Ni-rich (NiTi) and Ti-rich TiNi (TiNi) shape memory alloys were successfully fabricated by VPS process. Their phase transformation behavior, SME and SE, and mechanical properties were comparable to those of the bulk specimens reported by the previous researchers. The following is the summary of our findings on NiTi and TiNi samples. 312

4.1. NiTi sample (Ni-rich composition)

The NiTi sample in as-homogenized state did not exhibit 319 significant martensitic and reverse martensitic transformation 320 behavior as evidenced by the DSC study. The subsequent 321 aging at 773 K for 18 ks under air and vacuum condition 322 led to the excellent martensitic transformation behavior. The 323 aged NiTi sample tested at $M_{\rm f}$ and $A_{\rm f}$ showed SME and SE, 324 respectively, where the strain measured in the stress-strain 325 curves were more than 2.4% due to SME and 5.0% due 326 to SE, respectively. The thermal cyclic testing under con-327 stant stress was performed. The recoverable strain became 328 smaller with increasing number of cycle. After 15 cycles, the 329 strain-temperature curves were nearly saturated. The stress 330 cyclic testing was also conducted, where the permanent strain 331 was introduced during the testing. The plateau region due to 332 stress-induced martensite and its onset stress decreased with 333 increasing number of cycle. The recoverable strain after 200 334 cycles was 2.6%, which is useful as an actuator material The 335 TEM observation identified the formation of Ti₃Ni₄ precipi-336

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tates by the aging, responsible for increasing transformationtemperatures.

4.2. TiNi sample (Ti-rich composition)

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The TiNi sample in as-homogenized state showed clear trans-340 formation behavior as evidenced by the DSC measurements. 341 However, the aging treatment did not affect the transforma-342 tion temperatures. The TiNi sample in the as-homogenized 343 state exhibited SME behavior (4.0% recoverable strain) at RT. 344 The stress-strain curves were obtained at the $A_{\rm f}$, where no SE 345 behavior was observed. The TiNi sample has relatively high $A_{\rm f}$ 346 temperature, so that the degradation of mechanical properties 347 occurred at and above $A_{\rm f}$. At these temperatures, the sample was 348 yielded before the formation of stress-induced martensite. Dur-349 ing thermal cycle testing, the permanent strain due to the yielding 350 was introduced during first heating and thereafter. The recover-351 able strain decreased and the strain due to the yielding increased 352 with increasing number of cycle. Finally, the temperature-strain 353 curves are nearly saturated after 15 cycles. The recoverable 354 strain after 15 cycles was about 2.2%, which is also useful as an 355 actuator material. 356

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