

PHYS 575A/B/C

Autumn 2015

# Radiation and Radiation Detectors

Course home page:

<http://depts.washington.edu/phycert/radcert/575website/>

## 4: Scintillator counters; interactions of particles in matter

R. Jeffrey Wilkes

Department of Physics

B305 Physics-Astronomy Building

206-543-4232

[wilkes@u.washington.edu](mailto:wilkes@u.washington.edu)

# Course calendar

week	date	day	topic	text
1	10/1/15	Thurs	Introduction, review of basics, radioactivity, units for radiation and dosimetry	Ch. 1, notes
2	10/6/15	Tues	Radioactive sources; decay processes;	Ch. 1, notes
3	10/13/15	Tues	Photomultiplier tubes and scintillation counters; Counting statistics	Chs. 3, 8, 9 (I-V)
3	10/15/15	Thurs	<b>LAB: Room B248</b> Scopes, fast pulses; PMTs and scintillation counters; standard electronics modules	Chs. 4, 9, 16, 17
4	10/20/15	Tues	Overview of charged particle detectors	Ch. 4
4	10/22/15	Thurs	<b>LAB: Room B248</b> Coincidence techniques; nanosec time measurement, energy from pulse area	Chs. 17, 18
5	10/27/15	Tues	Interaction of charged particles and photons with matter	Ch. 2
6	11/3/15	Tues	Other photodetectors; gas and solid-state detectors	Chs. 5, 6, 7 Chs. 11, 12, 13
7	11/10/15	Tues	Detecting neutral particles; Data acquisition methods	Ch. 14, 15, 18
8	11/17/15	Tues	Cherenkov detectors; Case studies: neutrino detectors (Super-K)	Ch. 19, Notes
9	11/24/15	Tues	Case studies: classic detectors (cloud and bubble chambers, nuclear emulsion), high energy accelerators	Ch. 19
10	12/1/15	Tues	Case studies: contemporary leading-edge detectors (ATLAS, Auger)	Notes
11	12/8/15	Tues	Student presentations	-
11	12/10/15	Thurs	Student presentations	

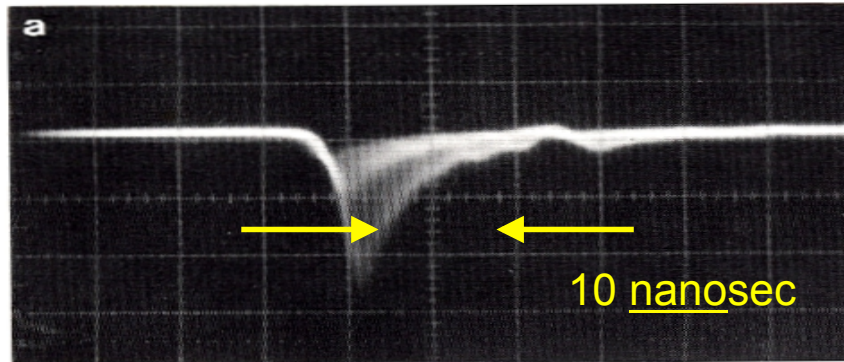
Tonight

# LAB session **this Thursday**

- Meet in room B-248, not here
  - 6:30 to 9pm
  - BEFORE class, read handouts posted on website:  
[“Documents for lab sessions”](#) (writeups and handouts)”  
[http://depts.washington.edu/physcert/radcert/575website/lab\\_documents/Lab\\_2/](http://depts.washington.edu/physcert/radcert/575website/lab_documents/Lab_2/)
1. Procedures for Lab 2: Energy Measurement Using PMTs and PHAs
  2. Labview pulse height analyzer software documentation
- Tonight:
    1. How scintillators work
    2. “Interactions of charged particles with matter” (energy loss processes in detectors and shielding)

Last time

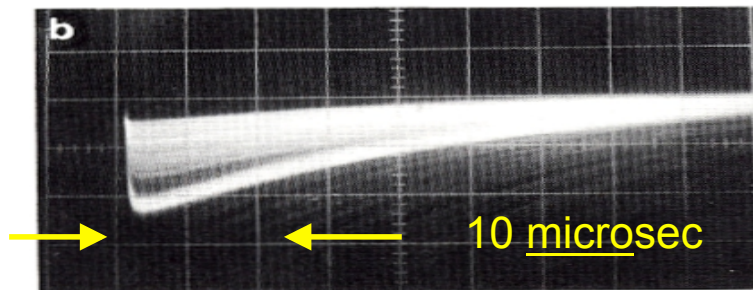
## Oscilloscope Traces from Scintillation Counters



### Plastic scintillator

Plastic  
Vert. scale : 0.2 V/cm  
Hor. scale : 10 ns/cm  
Source :  $^{207}\text{Bi}$  10 $\mu\text{Ci}$

10 nsec / division



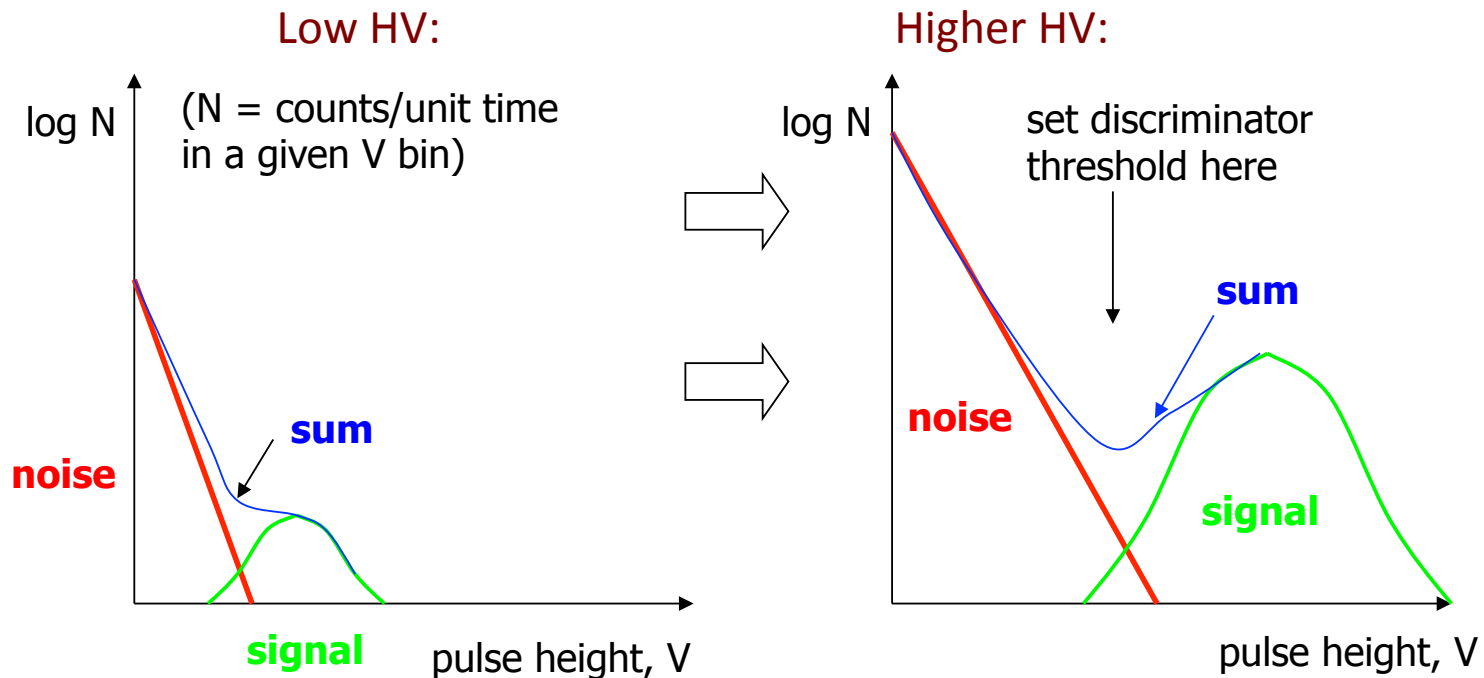
### Inorganic crystal, NaI

NaI  
Vert. scale : 0.2 V/cm  
Hor. scale : 5 $\mu\text{s}$ /cm  
Source :  $^{137}\text{Cs}$  10 $\mu\text{Ci}$

5000 nsec / division  
(Longer time scale for  
fluorescence to occur)

## Pulse height spectrum = histogram of pulse sizes

- Low amplitude pulses = (exponential) noise spectrum from PMT
  - Slope varies with temperature, photocathode material, tube gain
- Signal peak is at higher amplitude (pulse "height" = *area*)
- Whole pattern **expands** as  $V_{HV}$  is increased
- Goal: set discriminator threshold at "notch" between noise and signal.



Device to make these plots: Pulse Height Analyzer (PHA)

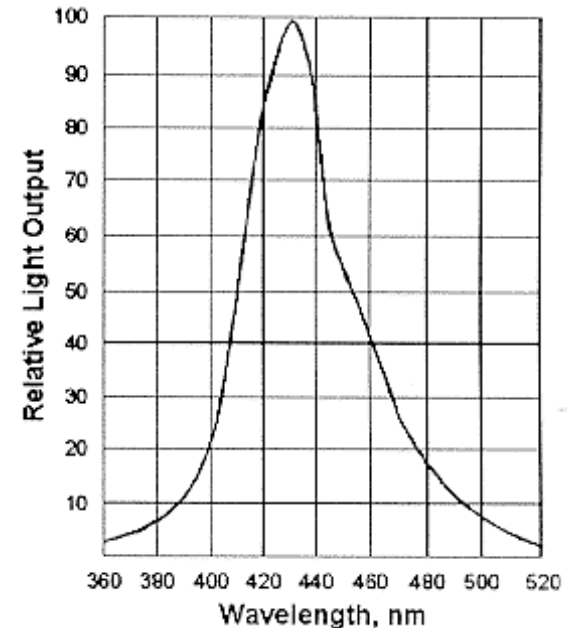
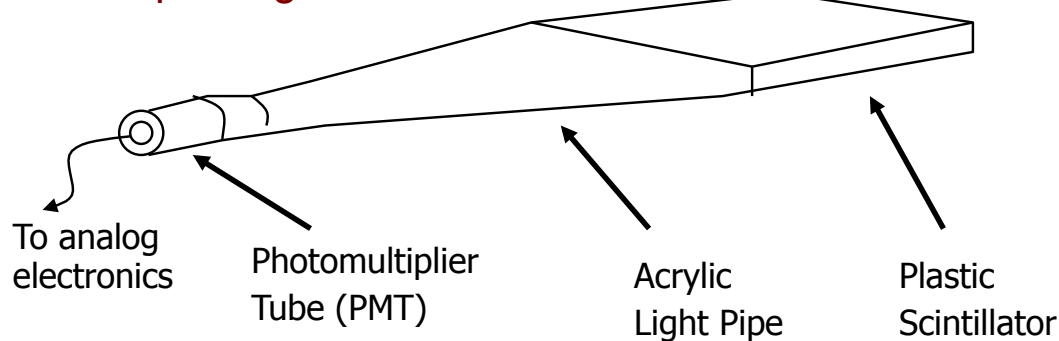
# Charged particle detectors: (1) scintillation counters

- Scintillators - features
  - Fast light pulses (nsec),
  - high yield ( $\sim 10^4$  photons per MeV of particle energy deposited),
  - emission at UV~blue wavelengths (400~450 nm)
    - Plastic scintillator = transparent styrene doped with fluorescent material
    - NaI = natural crystal with fluorescent behavior

Used in the form of strips, slabs, or fibers,  
with photons propagated by total internal  
reflection

Typical photon spectrum  
(Bicron Type 408)  
see <http://www.bicron.com/>

Example: big muon counter in our lab



# Scintillator examples

- Applications in Nuclear and Particle Physics experiments
  - Telescope or trigger
    - Defines **trajectory** of charged particles
    - Defines **time** when particle passed through scintillator and therefore can be used to start a clock for other detector elements
  - Total **Energy** Measurements
    - Particle stops in scintillator and deposits all of its energy
    - Use in sampling **calorimeters** where absorber is placed between scintillator planes, and charged particles created in the absorber deposit energy in the scintillator.
- Basic scintillation counter detector
  - **Scintillating substance** that emits light (solid, liquid, gas..)
    - Must be **transparent** to scintillation photons, and relatively inexpensive
    - Inorganic (crystals like NaI) or organic (plastic or mineral oil, with fluor)
  - **Sensitive light detector**
    - Photomultiplier (PM) tubes widely used
    - Silicon photodiodes (“solid state PMTs”)

## Scintillators: how they work

- Radiation interacts with atoms/molecules by **ionization**
  - leaves **ionization trail** that excites fluor atoms
- Light is emitted when atoms relax to ground states
- Light is collected and led to a photosensor
  - Photomultiplier tube
  - “solid state PMT” (SSPMT) = photodiode array with 1-photon sensitivity
- Scintillator counter can be used to measure **energy loss**
  - Energy loss by particle → light produced - **almost linear response**
- Fast response time – can be picosec, depending on material

<u>Property</u>	<u>Inorganic (crystal)</u>	<u>Organic (doped plastic)</u>
Track quenching	Small	Large
Time constant	Slow (~ $\mu$ s)	Fast (tens of ns)
Temperature dependence	Large	Small
Radiation Damage	Creation of long term trapping centers	Destruction of primary fluors
Density	Generally high, 3.67 NaI(Tl), 8.28 PbWO <sub>4</sub>	Always low, 1 g/cm <sup>3</sup> ~ CH <sub>2</sub>



# Scintillator materials

- Luminescence – absorbs energy and emits light in the visible spectrum
  - *Fluorescence* when emission occurs within  $10^{-8}$  s or so
  - Metastable states delay light emission leading to *phosphorescence* (light emission persists for  $10^{-4}$  s or more)
- Light emission often has two time constants, fast and slow
  - The number of photons can be described by sum of exponentials:  $N = N_0(e^{-t/\tau_1} + e^{-t/\tau_2})$  where  $\tau_1$  and  $\tau_2$  are the fast and slow components
    - Example: in BaF,  $\gamma$  rays excite the fast component, while  $\alpha$  particles excite the slow component: can identify them from the PMT signal.
- **Organic Scintillator**
  - Hydrocarbon compounds - nanosecond decay times
  - Organic crystals such as naphthalene ( $C_{10}H_8$ ) in a liquid organic solvent or a solid plastic solvent
  - Styrene with POPOP = 1, 4-bis(5-phenyloxazole-2-yl)benzene

# Desired attributes

- High efficiency to convert ionization energy to fluorescence
- Long attenuation length for emitted light
- Emitted light well matched to spectral response of photosensor, usually a PMT
- Fast decay time needed, when timing data is desired
- Average energy required per photon produced
  - NaI(Tl) – 25 eV/photon
  - Anthracene ( $C_{14}H_{10}$ ) – 60 eV/photon
  - Plastic + POPOP – 100 eV/photon
  - BGO ( $Be_4Ge_3O_{12}$ ) – 300 eV/photon

# Plastic scintillator is most common

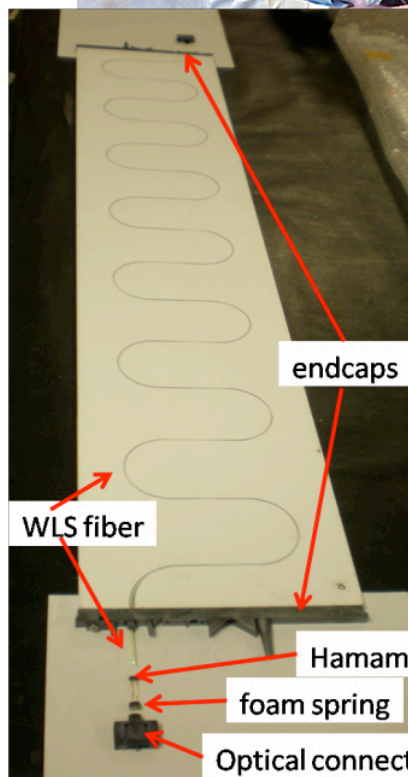
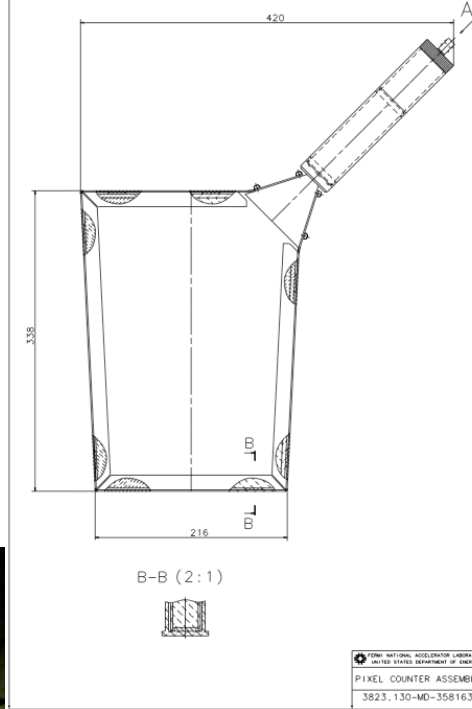
- Easy to work with
  - Can be cut to any shape
  - Can be bent into curves using solvents
  - Material not expensive
  - Can form by extrusion
  - Can machine plastic to achieve desired shape
- Fast signal (short decay time)
- Plastic light guides are easy to shape and attach by gluing with a plastic solvent; match optical properties
- Wave-length shifter can be added to plastic to shift the spectrum to longer wavelengths for more efficient response from silicon photosensors

## Different ways to get light out of a scintillator

Gluing PMTs onto scintillator faces



## Corner-mounted PMT



endcaps

WLS fiber

Hamamatsu MPPC

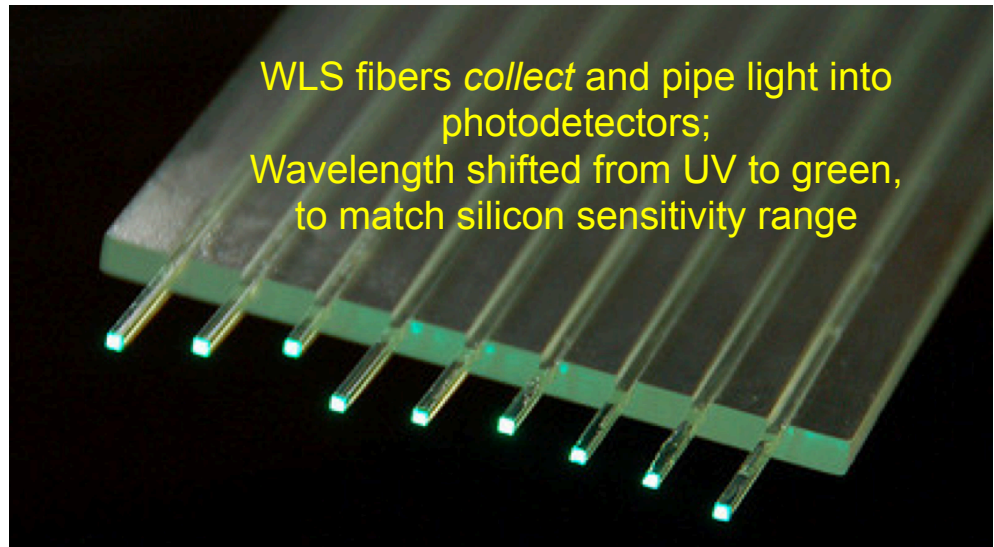
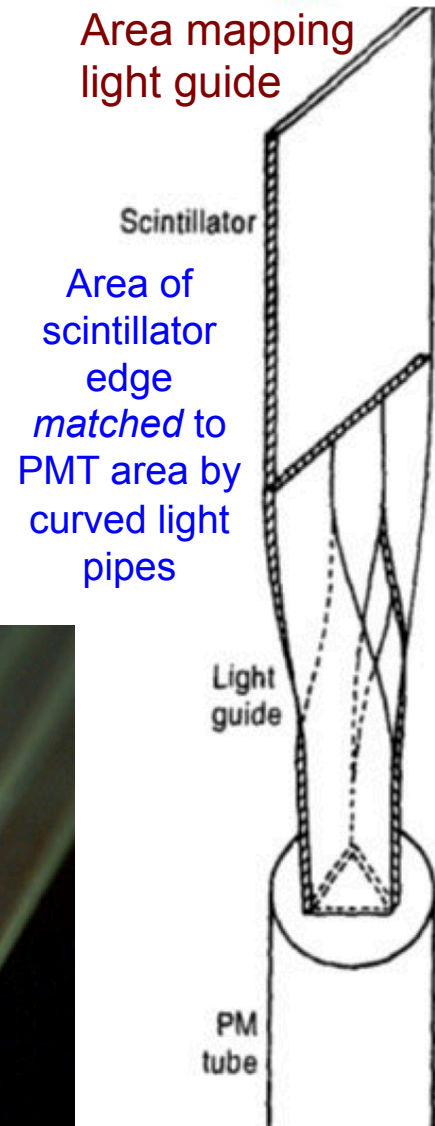
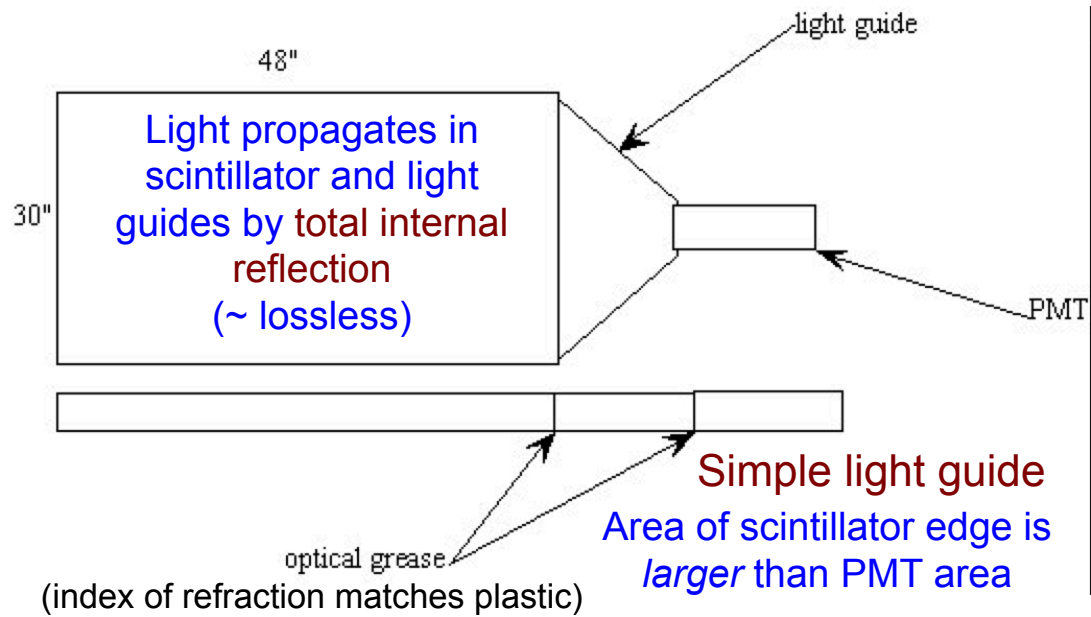
foam spring

Optical connector

Scintillator with Wavelength Shifting (WLS) optical fiber to collect and transport light to a photodiode

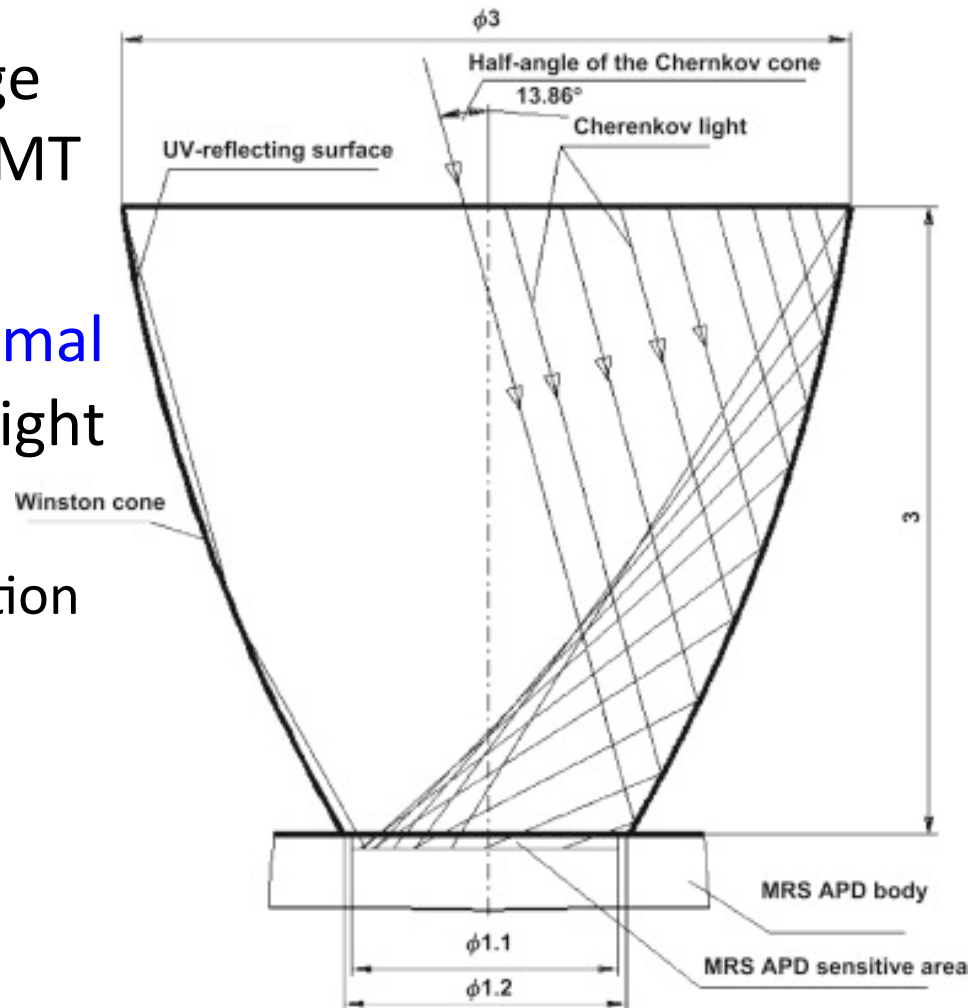


Lightproofing counters

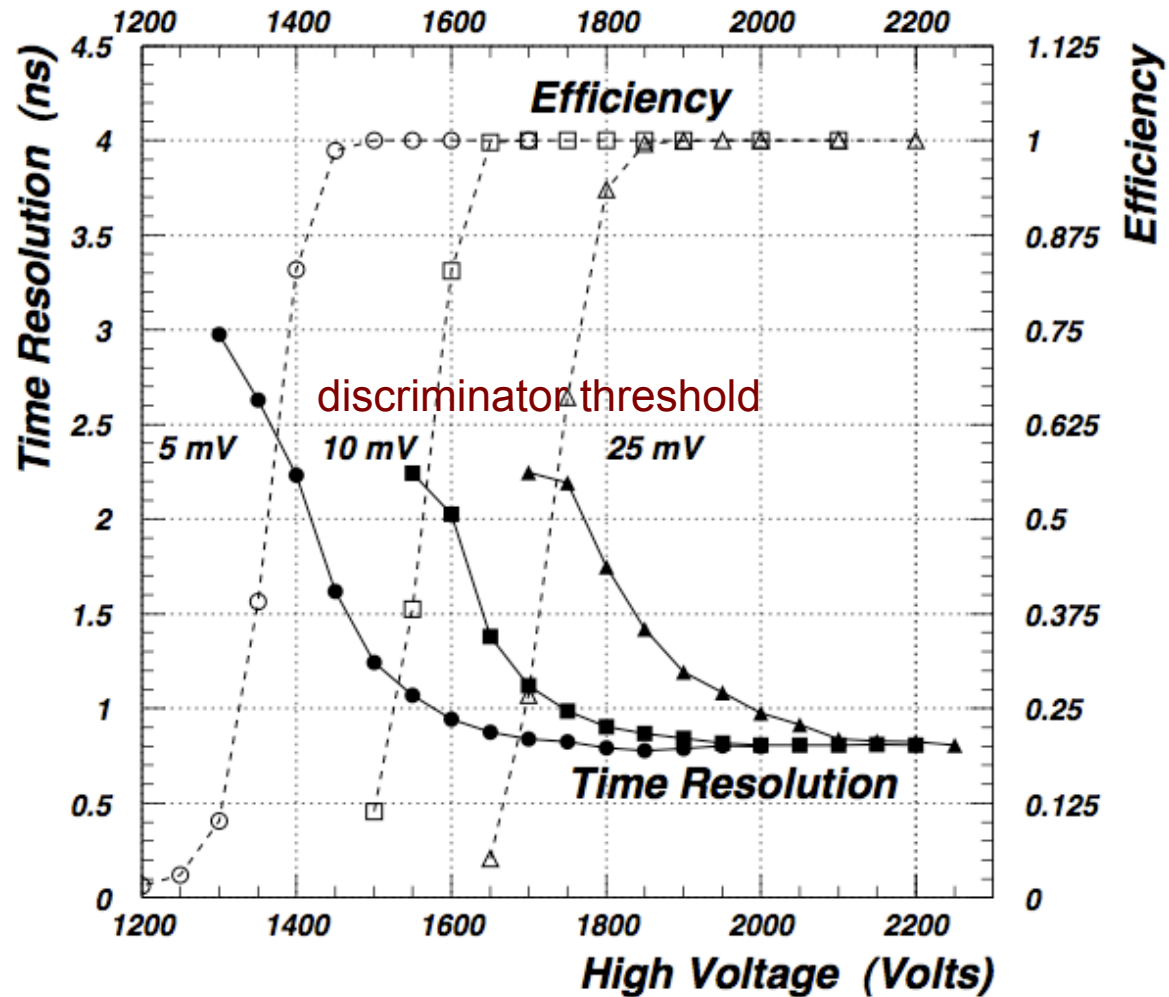


# Winston Cones

- Often scintillator edge area is bigger than PMT window area
- Winston Cone = **Optimal shape** for gathering light into a **smaller** area
  - Paraboloid of revolution



Example of decisions to be made setting up counters:  
Efficiency (fraction of tracks through counter that actually get counted)  
and time resolution vs HV and discriminator threshold voltage



# Particle interactions in matter

- Electromagnetic interactions

- Ionization
- Fluorescence
- Photoelectric effect
- Rayleigh and Raman scattering
- Compton effect
- Bremsstrahlung
- Pair production

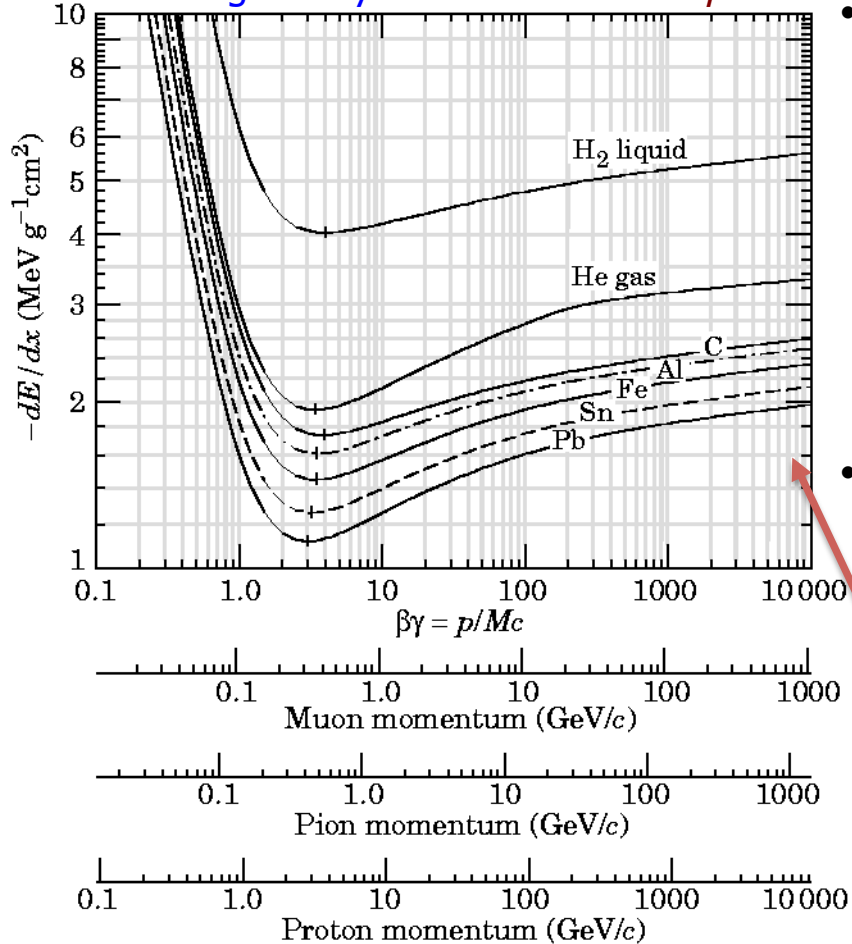


- Weak interactions
  - Radioactive decay
  - Neutrino interactions
- Strong interactions
  - Elastic scattering
  - Inelastic scattering
  - Fragmentation
  - Multiparticle production



# Electromagnetic interactions of particles in matter

Curves are given by the *Bethe-Bloch equation*\*



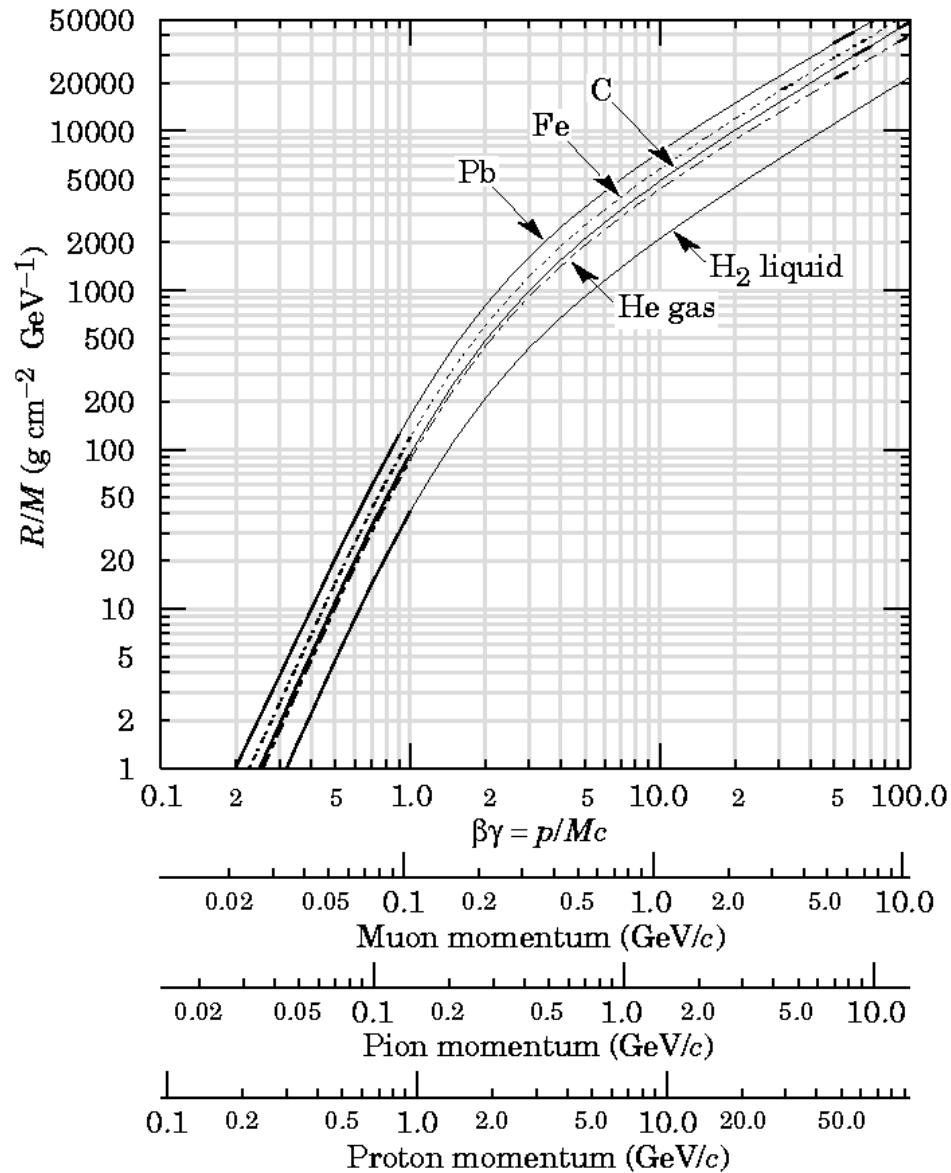
- Ionization
  - Charged particle leaves ionization trail via inelastic Coulomb interactions with atomic electrons
  - Basis for most particle detectors (scintillators, gas detectors, solid state, bubble and cloud chambers, nuclear emulsion, etc)
- Ionization energy loss ( $dE/dx$ ) or *stopping power*
  - Minimum-ionizing particles ( $\gamma = E/m \gg 1$ )

Rule of thumb:  
 $dE/dx \sim 2 \text{ MeV}/(\text{g}/\text{cm}^2)$   
 for “minimum-ionizing” particles  
 (actually: relativistic plateau)

$$* \quad -\frac{dE}{dx} = K z^2 \frac{Z}{A} \frac{1}{\beta^2} \left[ \frac{1}{2} \ln \frac{2m_e c^2 \beta^2 \gamma^2 T_{max}}{I^2} - \beta^2 - \frac{\delta}{2} \right]$$

See <http://pdg.lbl.gov/> for a wide variety of reviews and data tables

# Electromagnetic interactions of particles in matter



- **Range:** depth (in  $\text{g/cm}^2$ ) that a particle penetrates
  - Integrate  $dE/dx$  to get range (neglecting nuclear interactions)
  - Straggling: random fluctuations in range - few % for mass > electron

# Deflection processes

- **Multiple scattering** = Coulomb scattering off nuclei

$$\theta_{\text{plane}} = (1/\sqrt{2})\theta_{\text{space}} = Z(13.6 \text{ MeV}/\beta cp) \sqrt{(x/X_0) \{1+0.038 \ln(x/X_0)\}}$$

$x/X_0$  = thickness of material in **radiation lengths** (described later...)

$Z$ =charge of particle,  $p$ =momentum, MeV/c

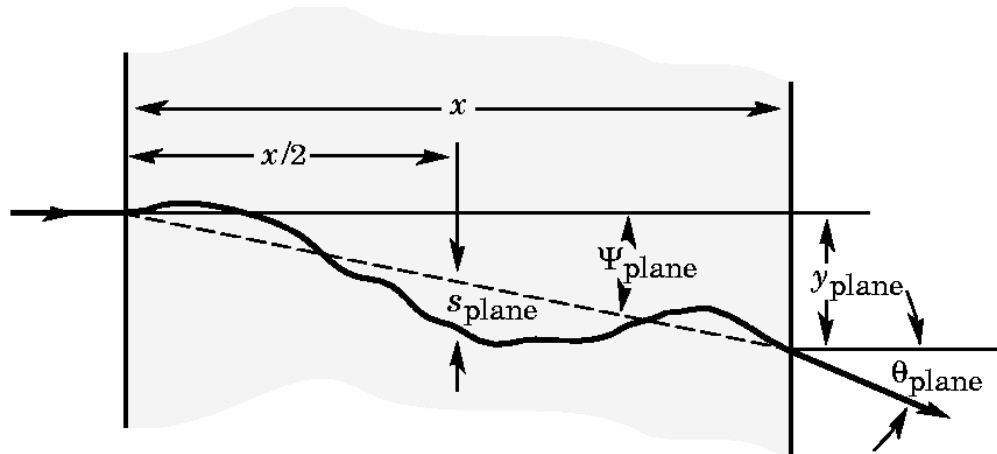
- **Magnetic deflection:**

Radius of curvature is

$$R(\text{cm}) = p(\text{MeV}) \frac{0.333}{Z \cdot B(\text{tesla})}$$

Note that  $R$  is proportional to  $p/Z$ , which is called the **rigidity**

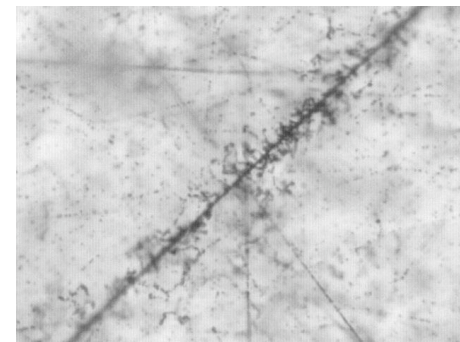
(rigidity units = MV, GV, etc).



- **Delta Rays**

- Energetic **knock-on electrons** from ionization events
- Tracks in detectors are mostly due to  $\delta$ -rays
- Can fake incoming particle tracks

Track image from nuclear emulsion plate exposed to cosmic rays by balloon flight in Antarctica (Wilkes, 1990) showing **heavy ion** track with delta rays (lighter tracks are minimum-ionizing protons)



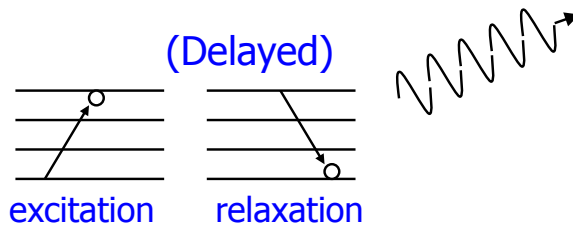
# Overview of photon-electron processes

- Initial state for all processes:

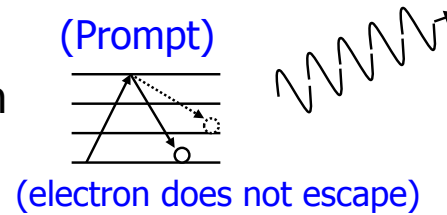


- Final states:

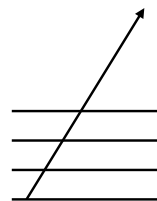
Fluorescence:



Rayleigh (elastic) or Raman (inelastic) scattering:

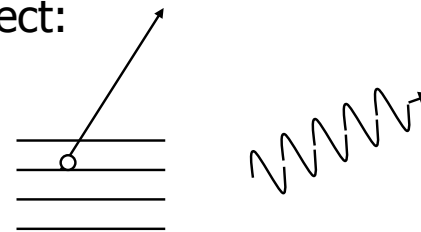


Photoelectric effect:



(free electrons produced)

Compton effect:



# What happens to **gamma rays** in matter

... depends on absorber material and its geometry

- **Photoelectric absorption** - atom absorbs gamma, e is ejected
  - Dominant for low E gammas ( $\ll 1$  MeV), or high Z absorber
  - Products can be
    - **Electron** with all the gamma's energy
    - In high-Z material: K shell electron ejected (10s of keV: L and M shells are usually too low-E to notice) – followed by atom's relaxation and another photon (**fluorescence x-rays**)
      - **Auger electron**: atom relaxes by emitting an **electron** instead of a fluorescence photon (much lower probability)
- **Compton electron** – gamma scatters *inelastically*
  - Process dominates in mid-range ( $\sim$ MeV) gamma energies
    - Atomic electron is kicked out with low kinetic energy – Compton electron
- **Pair production** – gamma disappears, becomes  $e^+ e^-$  pair
  - High energy gammas ( $\gg 1$  MeV) – in high-Z material
- **Coherent scattering**
  - elastic scattering, gamma just changes direction: **Rayleigh scattering**

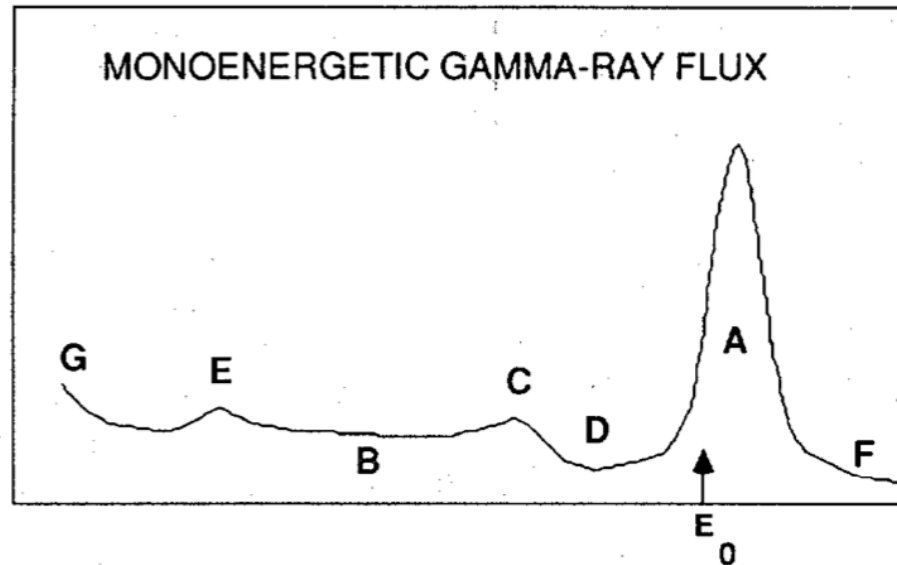
# Detectors for gamma rays

- Sodium Iodide NaI(Tl) is widely used for gamma ray spectroscopy
  - About 25eV/photon at the PMT = about four times more photons than we get from a plastic scintillator
  - But: **Hydroscopic**, must remain sealed! Easily ruined...
  - Light decay time is slow, about 230 ns
    - Not very good for fast timing applications
  - Has also some very long lived phosphorescence states
    - Mean decay time of 0.15 s
    - Amounts to about 9% of light yield
    - Can present problems in high counting rate applications
- Other Alkali Halide Scintillators
  - CsI(Tl) and CsI(Na) are often used in space applications because gamma-ray absorption coefficient per unit size is larger than NaI(Tl)

# What you see in the pulse height spectrum from a monoenergetic source

From: H. Smith and M. Lucas,  
*Gamma ray detectors*

**NUMBER OF PULSES**



**DETECTOR PULSE AMPLITUDE  
( $\gamma$ -RAY ENERGY)**

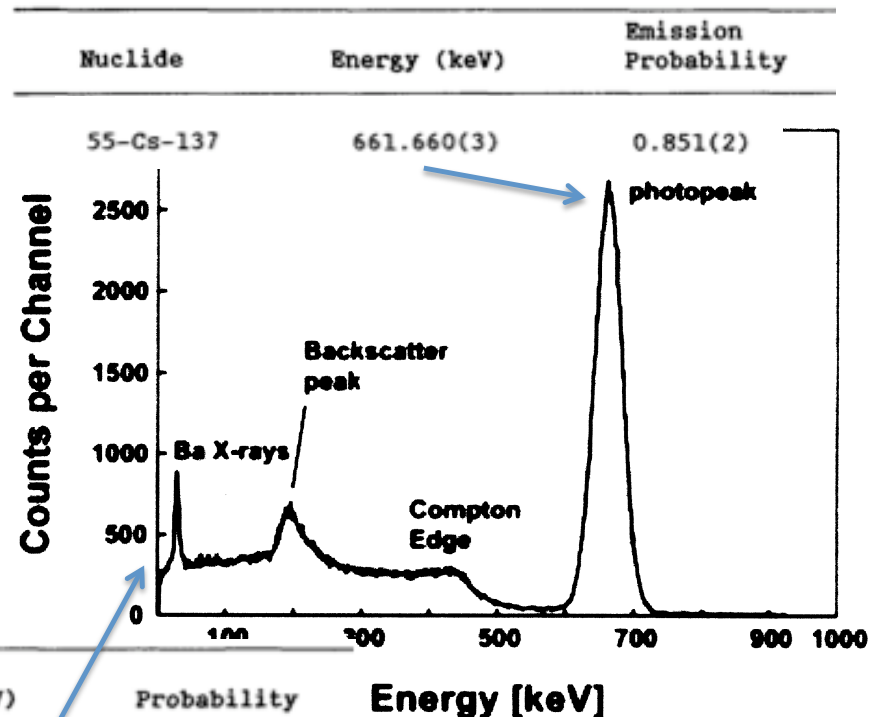
- A. Full-energy photopeak
- B. Compton continuum
- C. Compton edge
- D. Compton "valley"
- E. Backscatter peak

- F. Excess-energy region  
Background gammas
- G. Low-energy rise  
Noise in detector

# Example of photon interactions

- The light output of a scintillator such as NaI from gamma-rays emitted from a low energy source (less than a few MeV) will in general be a spectrum resulting from energy deposited by
  - Photoelectrons
  - Compton electrons
  - K-capture x-rays

*Typical pulse height spectrum of radiation emitted by a <sup>137</sup>Cs source detected in NaI(Tl)*



Nuclide	Trans	Energy (keV)	Probability
55-Cs-137	BaK $\alpha$	31.82-32.19	0.0566(16)
55-Cs-137	BaK $\beta$	36.36-37.45	0.0134(5)
55-Cs-137	BaK $\gamma$	31.82-37.45	0.0700(20)



# Understanding the observed pulse spectrum

- Energy spectrum of tracks detected in NaI from Cs-137
  - Betas are stopped by wrappings and not detected – only gammas
    - **But** gammas are not detected **directly** – only when they **create electrons**!
  - Features in the spectrum

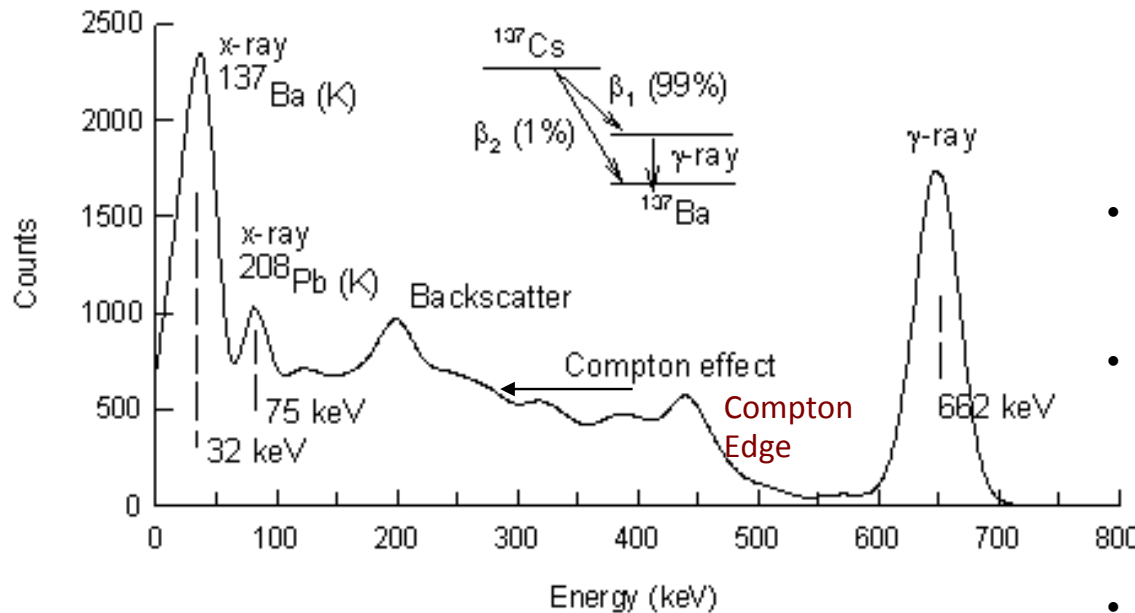
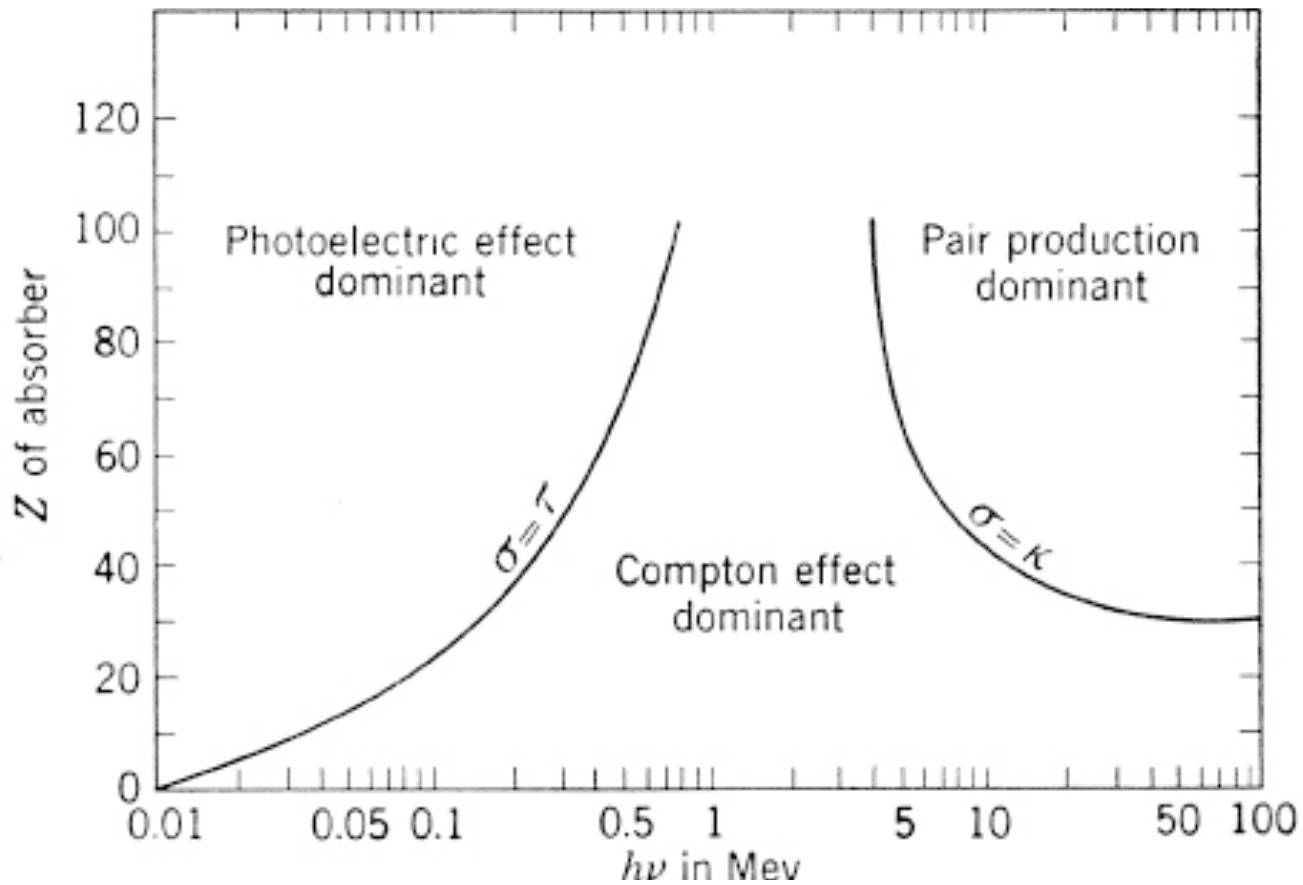


Figure 10.4 - Sample emission spectrum from  $^{137}\text{Cs}$ .  
From phys.fsu.edu

- Sharp lines  $\sim 30$  keV from de-excitation of daughter Ba nucleus via energy transfer from nucleus to inner-shell (K) atomic electrons
- “Photopeak” at 662 keV = full energy of gamma ray going to knocked out electron
- Compton effect: nearly flat spectrum from zero up to “Compton Edge” =  $(622 \text{ keV} - \text{max energy loss for gamma} = 181 \text{ keV for } 180\text{deg scatter}) = 481 \text{ keV}$
- Backscatter peak = gammas that Compton-scattered **outside** the counter and tend to have  $E \sim 181 \text{ keV}$  (=180 deg scatter)

# Energy and Z ranges for main processes

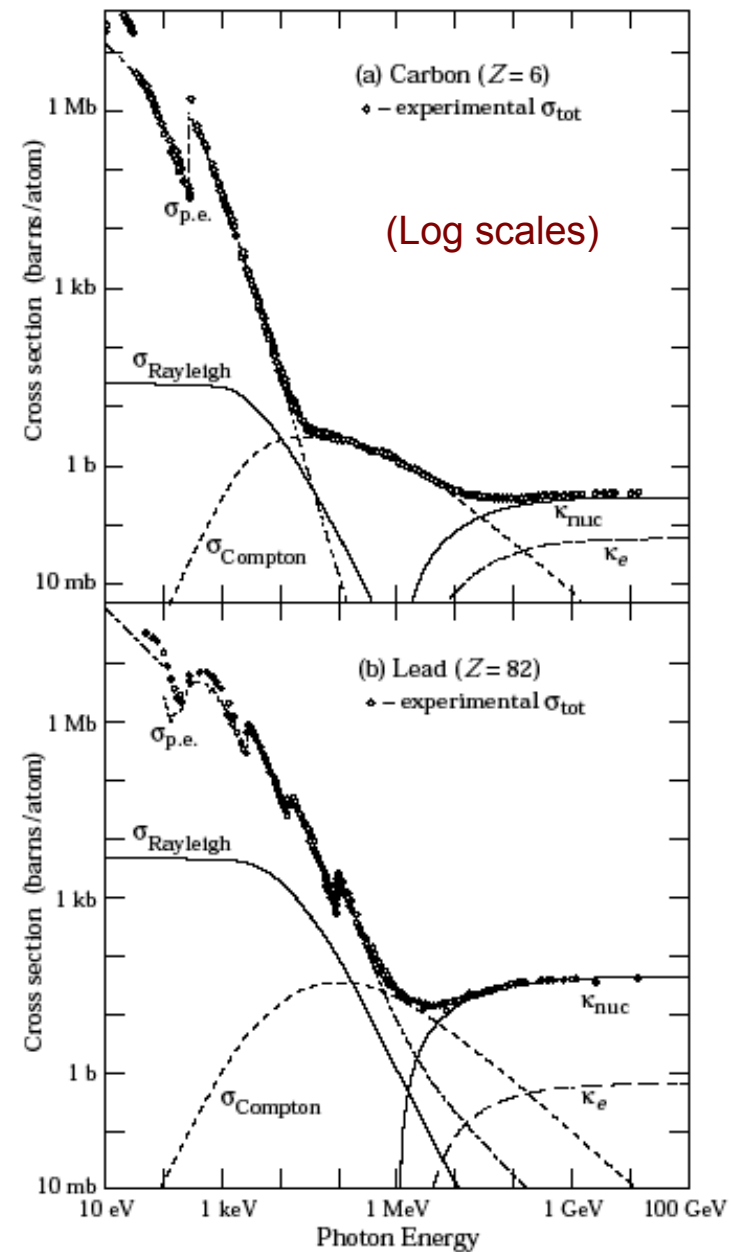


$$\text{Tau} \sim Z^4 / E^{3.5}$$

$$\text{Kappa} \sim Z^2$$

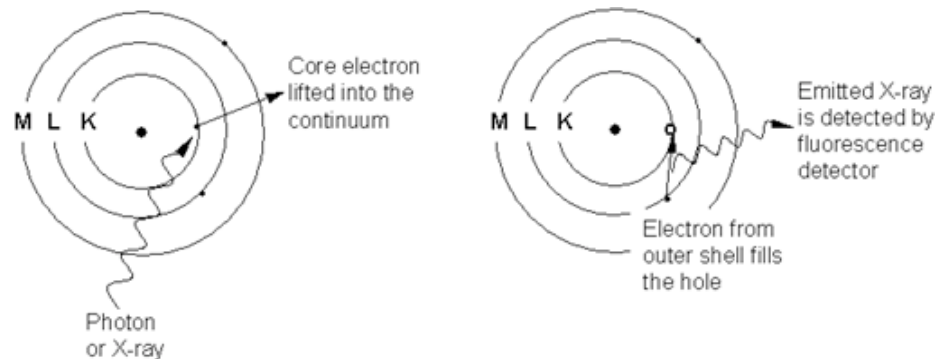
# Photon interactions in low and high Z

- For heavy nuclei, bremsstrahlung and pair production dominate earlier than for light nuclei
- low energies ( $< 100$  keV): Photoelectric effect
- medium energies ( $\sim 1$  MeV): Compton scattering
- high energies ( $> 10$  MeV):  $e^+e^-$  pair production



# Photoelectric absorption

- Dominant interaction mechanism for gamma rays < few MeV
- Gamma interacts with atom, and a *photoelectron* is ejected from one of the atom's *inner* (K, L) electron shells
  - Ejected photoelectron energy  $E_e = E_{\text{gamma}} - E_b$ , where  $E_b$  = binding energy of electron in its atomic state before interaction
  - Atom captures a free electron and/or rearranges its electrons shells
    - Typically one or more x-rays are emitted in this process
- The photoelectron carries essentially the entire energy of the gamma ray
  - Measurement of photoelectron's energy provides a good estimate of the original gamma ray's energy
  - If gammas come from a radioactive source, we may see a sharp “photopeak” in light output of scintillator detectors
    - Provides an estimate of  $Q$  = nuclear energy released in gamma decay



# Compton scattering

= Elastic collision of a gamma ray with an electron in absorbing material

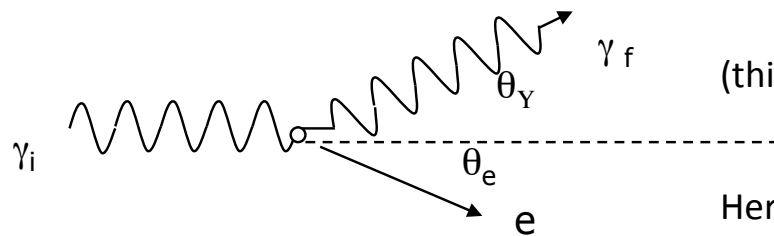
- The incoming gamma is scattered through an angle  $\theta$  with respect to its original direction
- A fraction of the photon's energy is transferred to the electron
  - recoils with energy ranging from 0 to a large fraction of the initial gamma ray energy
- From conservation of energy and momentum we obtain  
 $h\nu = h\nu' + h^2\nu^2 (1 - \cos\theta)/m_0c^2]$  and

$$\Delta\nu = \nu' - \nu = - \frac{2h\nu^2}{m_0c^2} \sin^2 \frac{\theta}{2}$$

where  $h\nu$  ( $h\nu'$ ) = energy of the incident (scattered) gamma ray and  $m_0c^2$  is the rest-mass energy of the electron

- This gives the maximum energy that can be transferred to the scattered electron - the scattered gamma ray never has 0 energy
  - Therefore the observed energy in, say, a Na-I scintillating detector will have a distribution with a cutoff – the Compton edge

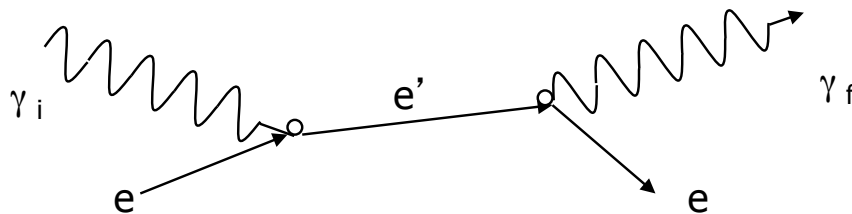
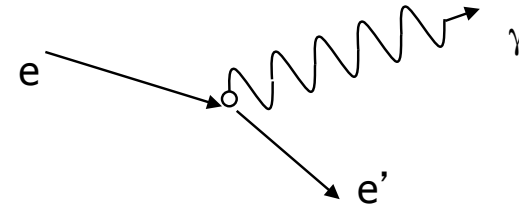
# Compton effect: inelastic photon-electron scattering



(this is just a sketch diagram of the process)

Here is a [Feynman space-time diagram](#). Time goes left to right, space up and down. Any photon-e interaction must be composed of [vertices](#) like this:

[Not just a sketch](#): Feynman's Quantum Electrodynamics (QED, 1948) associates parts of these diagrams with a mathematical factors in the cross section calculation for any EM process.



But if all 3 particles above are free particles, it is impossible to conserve energy-momentum.

Actual Compton process must go via a “virtual electron” (temporary violation of conservation of  $p+E$ , cancelled at 2nd vertex)

QED's “Feynman rules” associate a factor  $\alpha = e^2/(4\pi\epsilon_0\hbar c)$  ( $=1/137$ , dimensionless) with each vertex in the diagram, so the total cross section for Compton scattering should be on the order of  $\alpha^2$

# Compton kinematics

- Compton's **semi-classical** result (1923):

- Used **Planck's quantum hypothesis** ( $E=h\nu$ ) + **classical mechanics** ( $p=E/c$  for radiation, conservation of energy and momentum)

$$p_1 = E_1/c = h\nu_1/c = h/\lambda_1, \quad p_2 = h/\lambda_2$$

$E_1 \gg$  binding energy, so e is effectively free

- Conservation of  $\underline{p}$ :  $\underline{p}_e = \underline{p}_1 - \underline{p}_2$ , so  $p_e^2 = p_1^2 + p_2^2 - 2 p_1 p_2 \cos\theta_Y$
- Conservation of E:  $p_1 c + mc^2 = p_2 c + (mc^2 + p_e^2 c^2)^{1/2}$  (m=electron mass)

$$((p_1 - p_2)c + mc^2)^2 = (mc^2 + p_e^2 c^2)$$

$$p_e^2 = p_1^2 + p_2^2 - 2 p_1 p_2 + 2mc^2(p_1 - p_2)/c$$

- Equate RHS of these equations:

$$p_1^2 + p_2^2 - 2 p_1 p_2 \cos\theta_Y = p_1^2 + p_2^2 - 2 p_1 p_2 + 2mc^2(p_1 - p_2)/c$$

$$p_1 p_2 (1 - \cos\theta_Y) = mc^2(p_1 - p_2)/c$$

$$\text{multiply by } hc/(p_1 p_2 mc^2): (1 - \cos\theta_Y)hc/mc^2 = h/p_2 - h/p_1 = \lambda_2 - \lambda_1$$

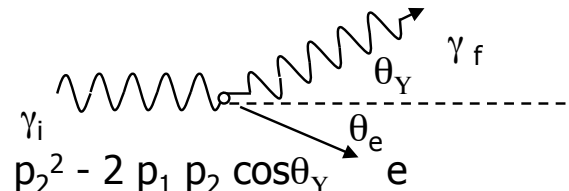
$hc/mc^2$  has **dimensions of length**:  $= \lambda_C$ , the **Compton wavelength of the electron**

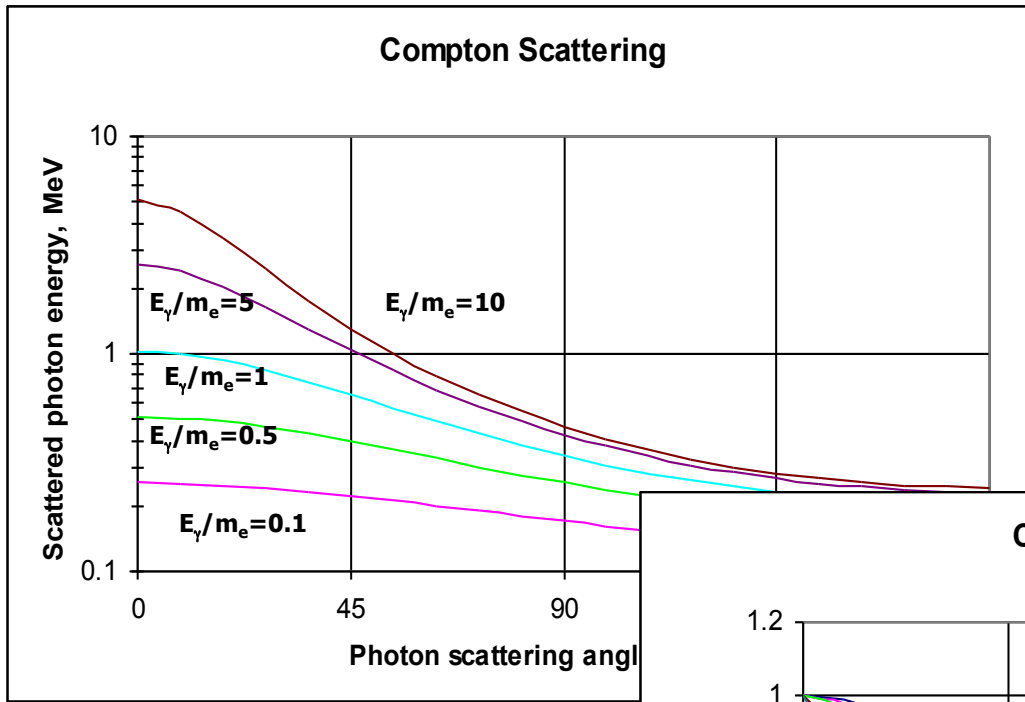
$$\lambda_C = 0.02 \text{ A (angstrom } \text{A}=10^{-10} \text{ m, typical atomic diameter)}$$

- **Result: change in wavelength (energy loss) of Compton scattered photon is given by**  $\lambda_2 - \lambda_1 = (1 - \cos\theta_Y) \lambda_C$

Using  $E=hc/\lambda$ , we can recast this as  $hc(1/E_2 - 1/E_1) = (1 - \cos\theta_Y) hc/mc^2$

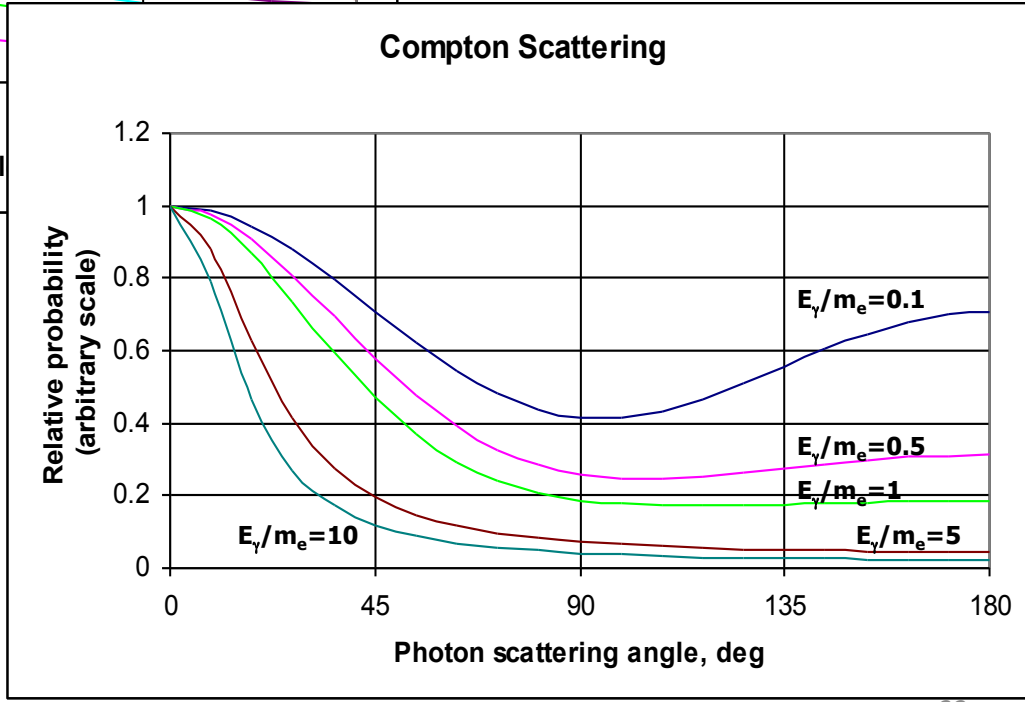
$$(1/E_2 - 1/E_1) = (1 - \cos\theta_Y)/mc^2$$





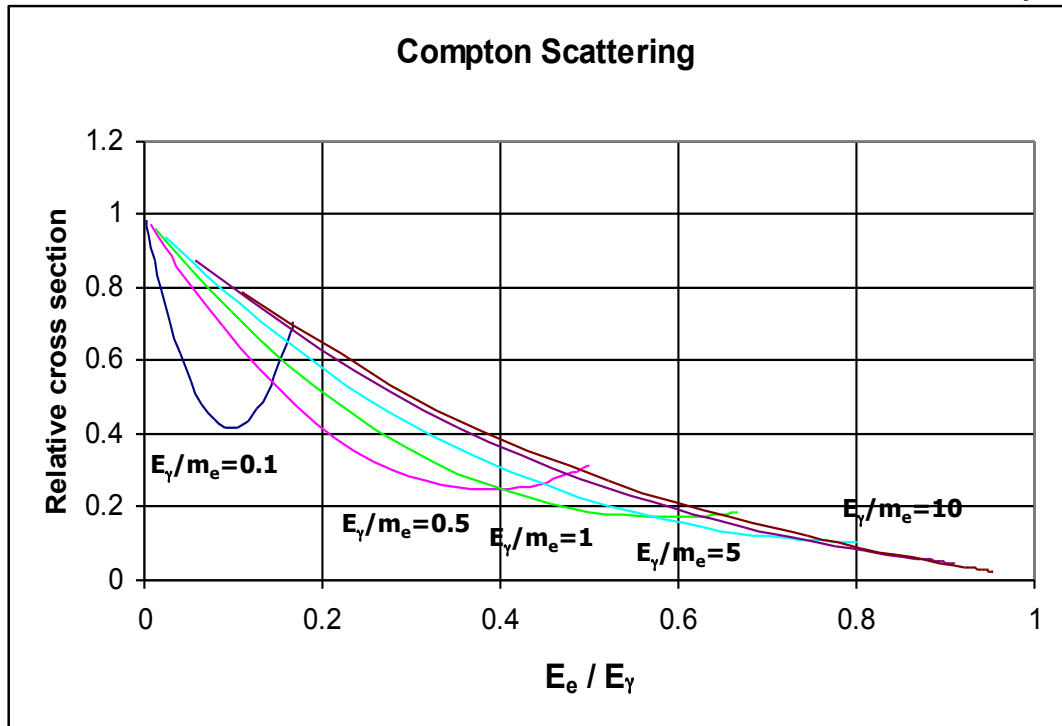
$dn/d\theta$  vs  $\theta$

and  $E_\gamma$  vs  $\theta$



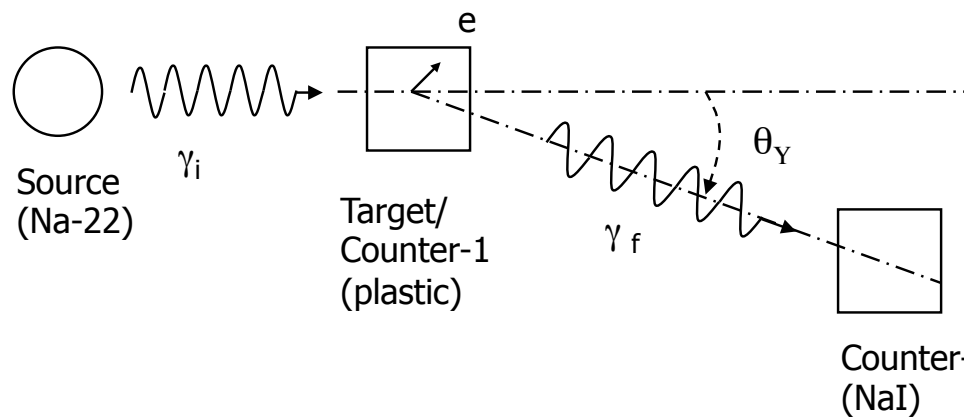


# Compton electron energy distribution



Relative probability of Compton scattering vs energy (in units of  $m_e$ )

Compton scattering detector setup:  
 Trigger: C1 .and. C2  
 Data: C2 pulse height spectrum, and total count vs angle



Note:  
 if C1 is too thick, it will stop  $\theta_Y$   
 if C2 is too small it will not absorb  $\theta_Y$

## Electromagnetic cascade processes

- **Bremsstrahlung**

(“braking radiation”) = emission of photon by charged particle when accelerated by electric field of a nucleus (or atomic electron)



High-energy electrons and photons create “cascades” (or “showers”) in dense materials, due to interplay between two processes: bremsstrahlung and pair production

- **Radiation length  $X_0$**

= distance ( $\text{g}/\text{cm}^2$ ) for 37% ( $1/e$ ) energy loss due to bremsstrahlung

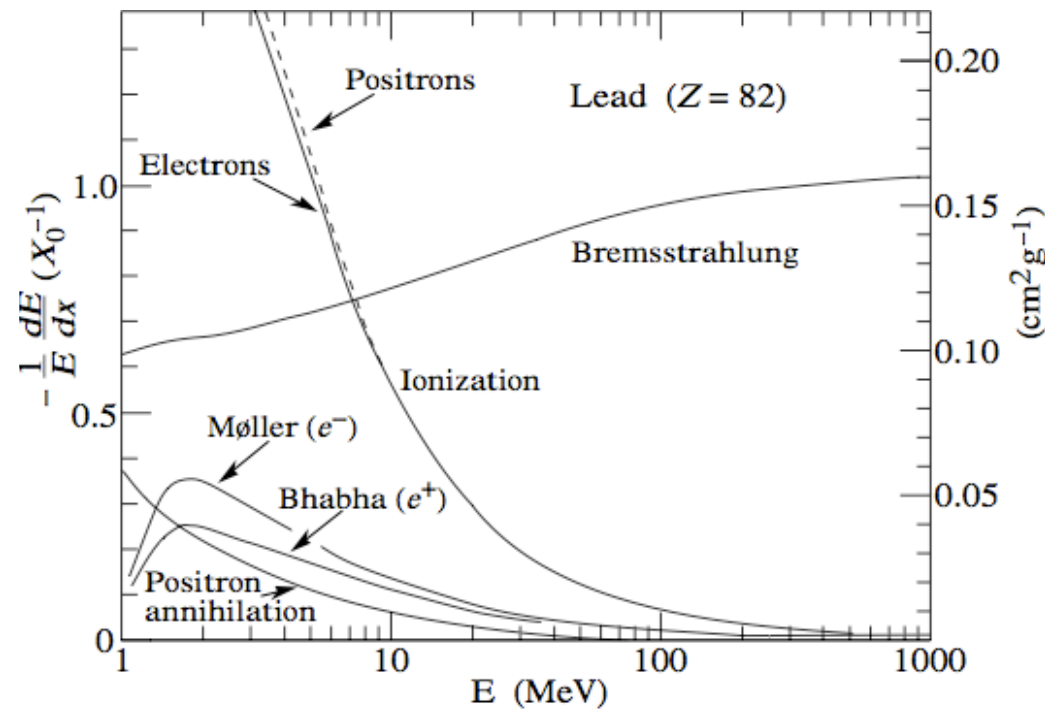
$$X_0 = \frac{716.4 \cdot A}{Z(Z + 1) \ln \frac{287}{\sqrt{Z}}} \text{ g} \cdot \text{cm}^{-2}$$

eg,  $X_0 =$

44  $\text{g}/\text{cm}^2$  for scintillator,

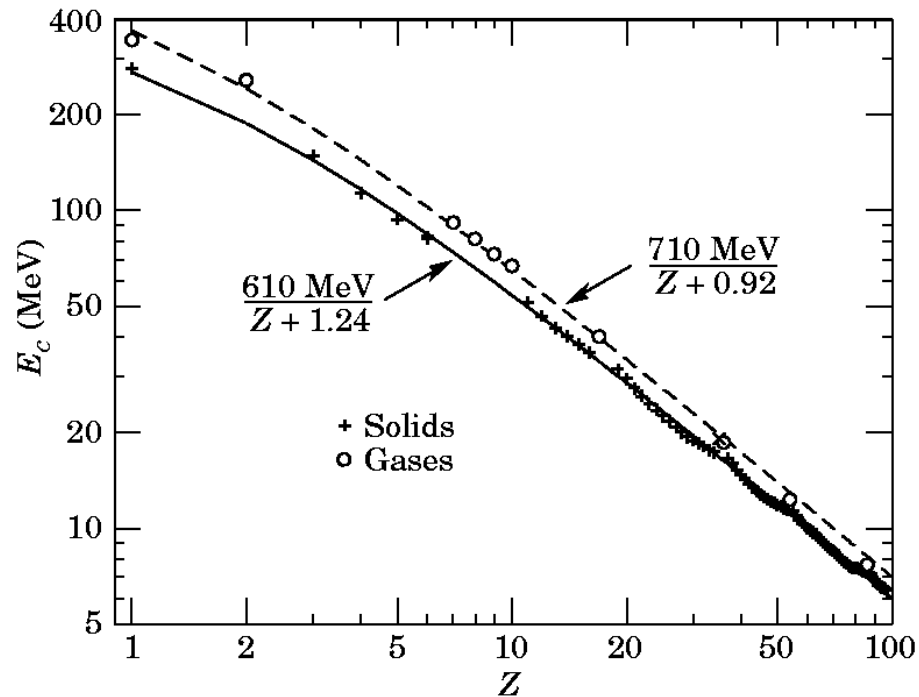
13.8  $\text{g}/\text{cm}^2$  for Fe,

6.4  $\text{g}/\text{cm}^2$  for Pb



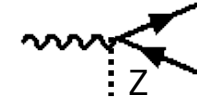
# Critical energy for materials

- **Critical energy** = energy at which  $dE/dx$  due to ionization is same as  $dE/dx$  due to bremsstrahlung
  - Below  $E_c$ , energy loss is predominantly due to ionization, above  $E_c$ , bremsstrahlung



# Pair production

Dominant energy loss mode for very high energy electrons;



For energies below 10 MeV it is negligible

- Possible for photon energies greater than 2X mass of the electron (1.02 MeV)
  - $\gamma$ -ray becomes an electron-positron pair in the E field of a nucleus
    - Particles' kinetic energy = initial photon E minus the rest-mass required to create the  $e^+e^-$  pair

Observed energy spectrum is complicated:

- The electrons lose energy by ionization or brem;
- The positron annihilates with an electron (after coming to rest) -> two  $\gamma$ -rays.

One or both gamma rays may escape the detector

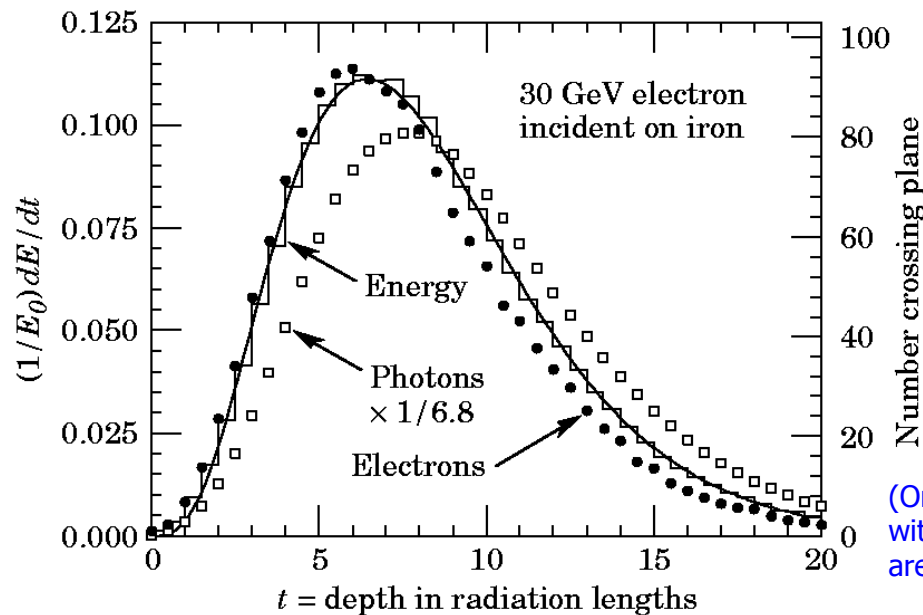
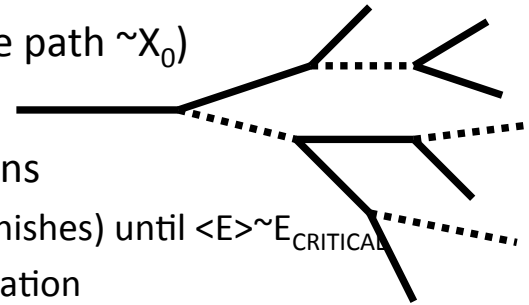
Result is three peaks in the energy spectrum:

- **total-energy peak** (E),
- **one-escape peak** ( $E - m_e c^2$ ) and a
- **two-escape peak** ( $E - 2m_e c^2$ )

For high energy photons or electrons, pair production and subsequent Bremsstrahlung radiation create more  $\gamma$ -rays, which in turn create more  $e^+e^-$  pairs etc. leading to a shower of electrons

# Electromagnetic cascade development

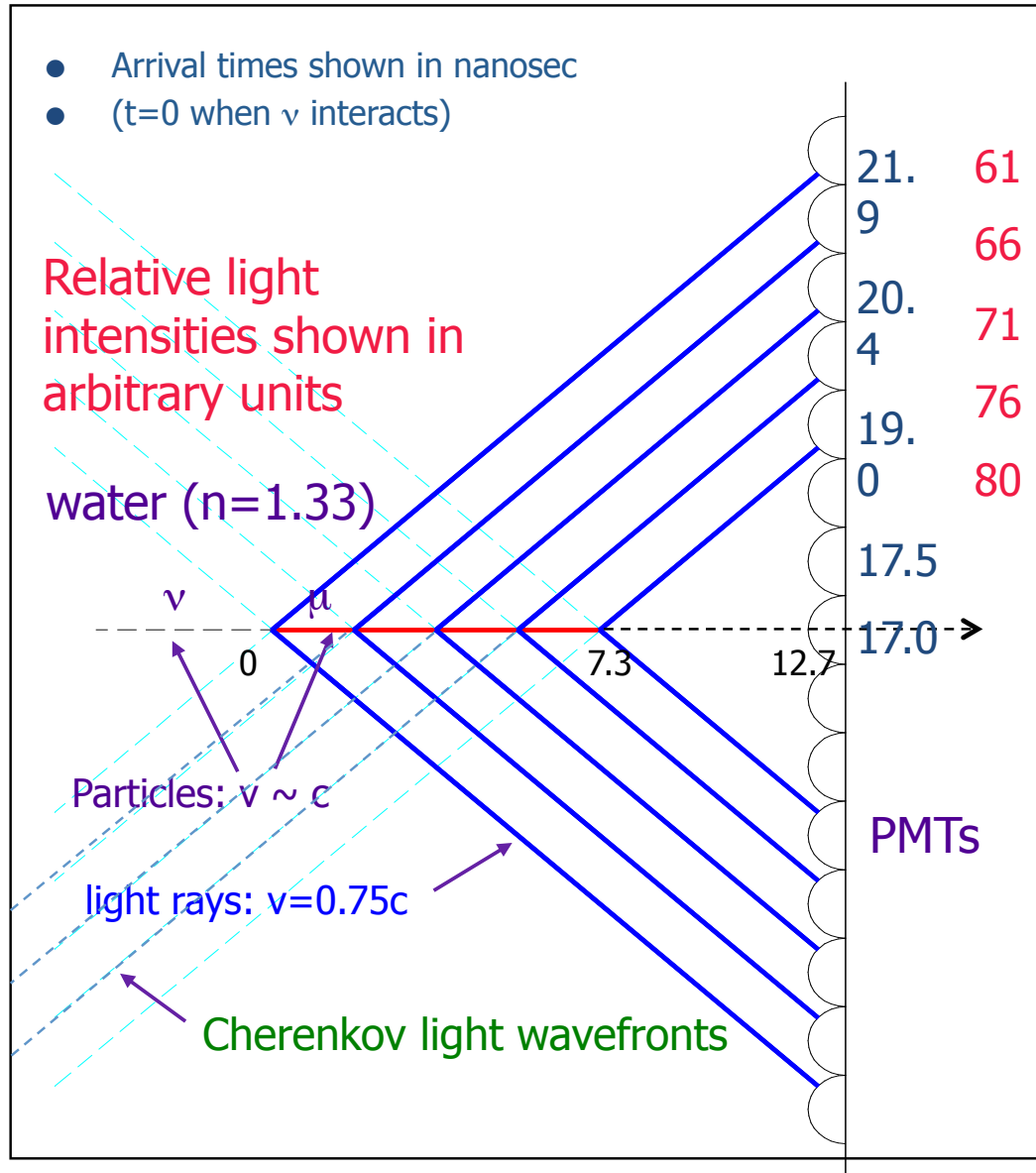
- Electron entering dense matter soon brems (mean free path  $\sim X_0$ )
- Brem photon soon pair produces (mfp  $\sim (7/9)X_0$ )
  - etc, etc: result is a *cascade* of electrons and photons
  - Number of particles builds up (and  $\langle E \rangle$  per particle diminishes) until  $\langle E \rangle \sim E_{\text{CRITICAL}}$ 
    - Then brems losses become less important than ionization



- Notice brems/PP cascade process does not *dissipate* energy, just swaps it from  $e'$ 's to photons and back again
- Main effect: *divide energy up* among more and more particles
- Energy is *lost to medium (heating)* *only* via ionization, after  $e'$ 's drop below  $E_c$

(Only particles with  $E > 1.5$  MeV are counted)

# Cherenkov radiation



- Charged particle may travel faster than the speed of light in a material medium
  - $(v_{E-M}=c/n)$ .
- Particle's field produces electromagnetic analog of a sonic boom

Applications:

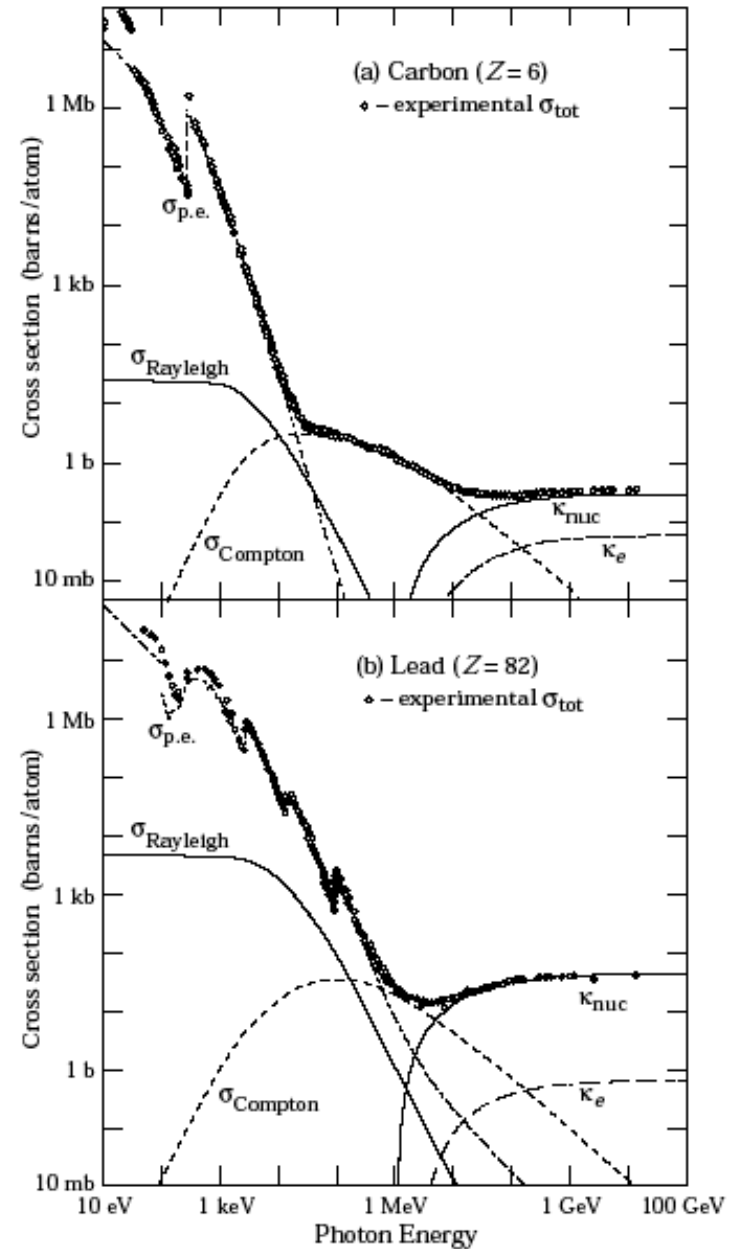
- Ring-imaging Water Cherenkov Detectors (Super-Kamiokande)
  - Reconstruct track
- Threshold detectors
  - Select particles faster or slower than  $v$
  - If momentum is known, identify mass

(shown earlier)

## Photon interactions overview

- For heavy nuclei, brem/pp dominate earlier than for light

- low energies ( $< 100$  keV):  
Photoelectric effect
- medium energies ( $\sim 1$  MeV):  
Compton scattering
- high energies ( $> 10$  MeV):  
 $e^+e^-$  pair production



# Cross sections

- First:  $\Omega$  **solid angle acceptance** of detector:  
Project detector area as viewed from source onto a 1m radius sphere:

- **Total cross section**  $\sigma = N_{\text{events}} / (\Delta t N_{\text{tgt}} I_{\text{beam}})$

$$I_{\text{beam}} = N_{\text{beam}} / \Delta t \Delta S \text{ for beam area } \Delta S \text{ larger than target}$$

$$N_{\text{tgt}} = \text{number of target nuclei in beam area}$$

(= number of electrons in beam for Compton scattering)

$$= (N_{\text{avogadro}} M_{\text{tgt}} / A_{\text{tgt}}) Z_{\text{tgt}} \quad (M_{\text{tgt}} = \text{moles})$$

$\sigma$  has dimensions of area, unit : **barn** =  $10^{-28} \text{ m}^2$

- Can have various **partial cross sections**:

- vs energy =  $d\sigma/dE$
- vs angle =  $d\sigma/d\theta$
- vs several things at once:  $d^3\sigma / dE d\theta d\Omega$   
= cross section at energy  $E$  (per unit  $E$ ) and angle  $\theta$  (per unit  $\theta$ ), per unit solid angle  $d\Omega$  centered on  $\theta$
- area under  $d\sigma/d(\text{whatever})$  curve =  $\sigma$

