

#### Improving Adhesive Bonding of Composites Through Surface Characterization

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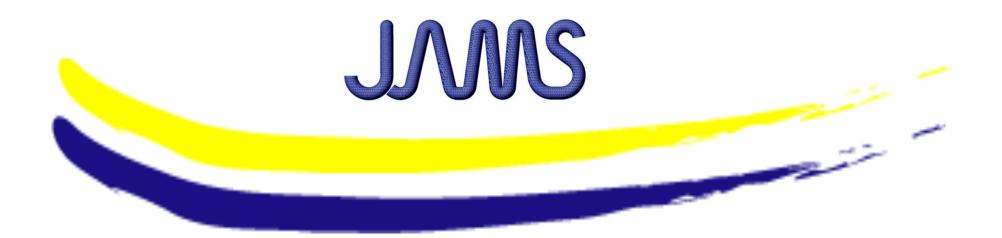








- Motivation and Key Issues
  - Adhesive bonding is being used for primary composite structure in commercial transport aircraft manufacture and repair- surface preparation is a critical step
  - Good bonds are produced but questions remain:
    - What are appropriate techniques to inspect surfaces?
    - What are key factors for making a good/poor bond?
    - How to predict material and surface preparation compatibility?
- Objective
  - Further understand the requirements for surface preparation to produce strong primary structural composite bonds with different substrates and adhesives



# Variables that affect contact angle measurements on peel ply surfaces

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#### VARIABLES THAT AFFECT CONTACT JMS ANGLE MEASUREMENTS ON PEEL PLY SURFACEs





- Motivation and Key Issues
  - Most important step for bonding is SURFACE PREPARATION!!
  - Inspect the surface prior to bonding to ensure proper surface preparation
- Objective
  - Develop QA technique for surface preparation
- Approach
  - Investigate variables that affect contact angle measurements
  - Verify technique on intentionally contaminate surfaces



### FAA Sponsored Project Information





- Principal Investigators & Researchers
  - Brian D. Flinn (PI)
  - Ashley Tracey (new PhD student, UW-MSE)
  - Jeffery Saterwhite (MS 2009 UW-MSE)
- FAA Technical Monitor
  - David Westlund
- Other FAA Personnel Involved
  - Larry Ilcewicz
- Industry Participation
  - Toray Composites
  - Henkel International
  - Precision Fabrics & Richmond Aerospace & Airtech International
  - The Boeing Company (Kay Blohowiak, Peter Van Voast, and William Grace)



- Surface energy and bonding
- Effect of time to measure contact angle on measurement
- Effect of peel ply orientation on contact angle measurement
- Detection of Si contamination
- Conclusions



- Why use surface energy to probe the surface preparation method applied to the composite for bonding?
  - One requirement of adhesion is the adhesive must wet the substrate
    - This is controlled by surface energy

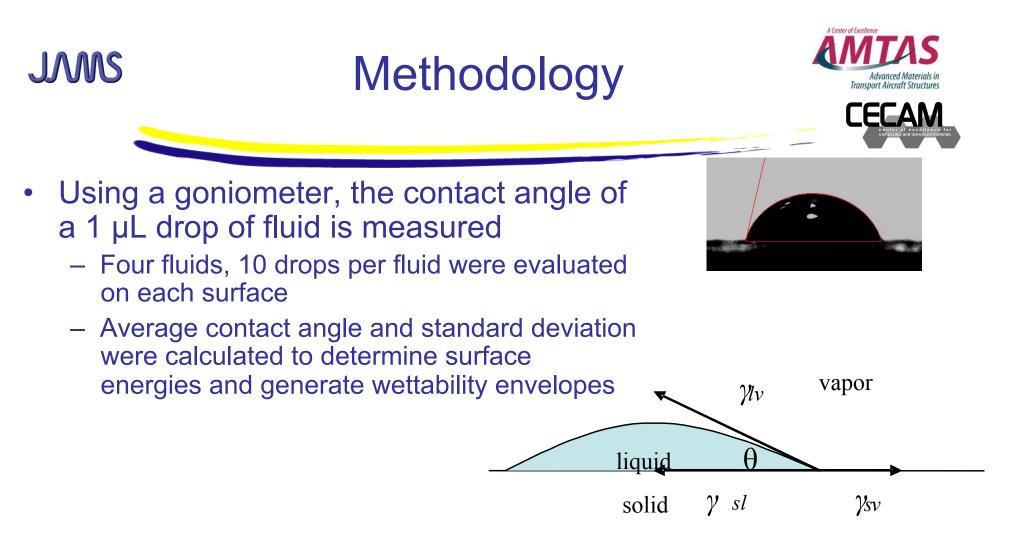




- Surface energy is a complex property composed of many components
  - Owen Wendt model breaks the surface energy into polar and dispersive components

 $\gamma_{\text{Total}} = \gamma_{\text{Polar}} + \gamma_{\text{Dispersive}}$ 

• Use this approach to probe the surface to determine if we can detect variations, or contamination that correlate with bond quality



- Complete wetting when θ approaches zero
- Contaminants usually lower the solid's surface energy (increase  $\theta$ )
- Surface preparations try to increase the solid's surface energy and clean off contaminants

### JMS Calculation of Surface Energy





• To calculate the surface energy using, the following equation was used:

$$\frac{\gamma_{lv}(\cos\theta+1)}{2\sqrt{\gamma_{lv}^d}} = \sqrt{\gamma_{sv}^p} \left(\sqrt{\frac{\gamma_{lv}^p}{\gamma_{lv}^d}}\right) + \sqrt{\gamma_{sv}^d}$$

- $\gamma_{lv}$  is the total surface energy between the liquid and the vapor,
- γ<sup>p</sup><sub>lv</sub> is the polar component of the surface energy between the liquid and vapor, γ<sup>d</sup><sub>lv</sub> is the dispersive component of the surface energy between the liquid and the vapor,
- γ<sup>p</sup><sub>sv</sub> is the polar component of the surface energy between the solid and the vapor,
- γ<sup>d</sup><sub>sv</sub> is the dispersive component of the surface energy between the solid and vapor, and
- θ is the average contact angle.



From equation (1), the following equations were used to plot the data:

$$\frac{\gamma_{lv}(\cos\theta+1)}{2\sqrt{\gamma_{lv}^d}} \quad (2) \qquad \sqrt{\frac{\gamma_{lv}^p}{\gamma_{lv}^d}} \quad (3)$$

- Where (2) is the y-coordinate and (3) is the x-coordinate
- From this plot (Kaelble plot), the polar and dispersive components of the surface energy are determined as follows.
  - Polar component = b<sup>2</sup> (b = y-intercept of plot)
    Dispersive component = m<sup>2</sup> (m = slope of

(m = slope of plot)

By inputting these polar and dispersive components of the surface energy of the composite into the computer program BKCWet v 1.1, wettability plots were generated.



- Toray 3900/T800 unidirectional laminates
- Precision Fabric Group 60001 polyester peel ply
- Autoclave cure of composite (max 176.7 °C, 0.6 MPa)
- Peel ply removed and contact angles measured within 1 hour
- Fluids used for contact angle analysis:
  - De-ionized water (DI water)
  - Dimethlysufoxide (DMSO)
  - Ethylene Glycol (EG)
  - Glycerol (Gly)
  - Formamide (Form)

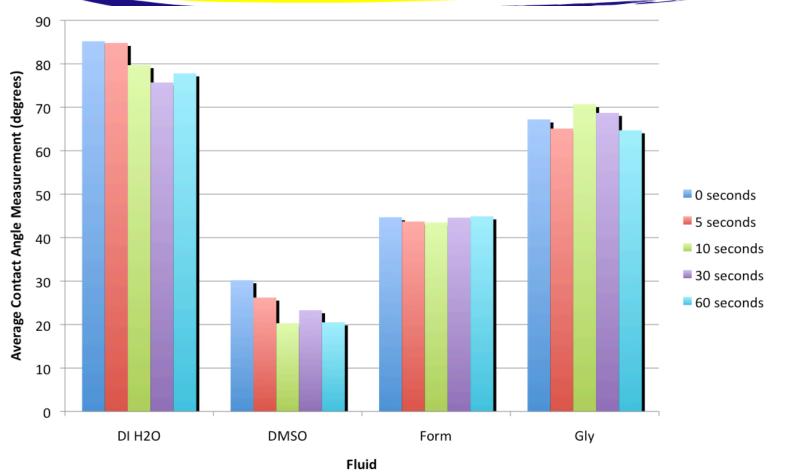


- Does the time at which the contact angle is measured after application of the liquid droplet to the solid surface effect the measurement?
  - Measure contact angles at 0+, 5, 10, 30 and 60 second
  - -4 different fluids

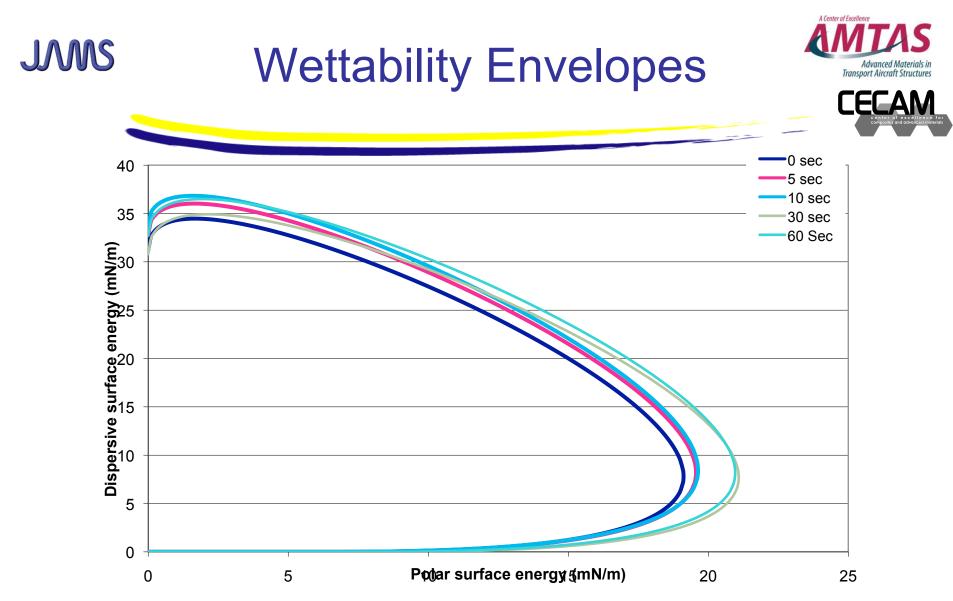
## JMS Effect of Time to Measure on Contact Angle (CA)







• DI H<sub>2</sub>O and DMSO contact angles decrease with increasing time



### Wettability envelopes increase with increasing time (contact angle decreases with time)



### Effect of Time on CA

120





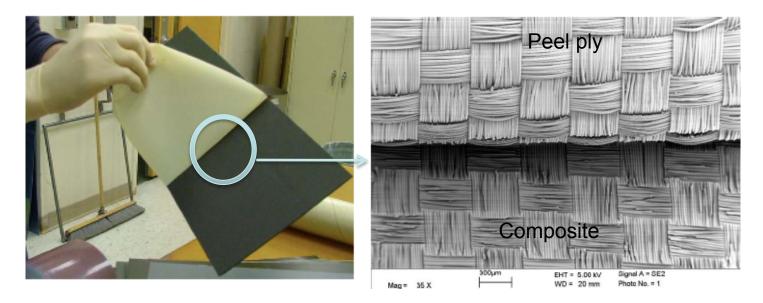
- To determine if this was a viscosity effect, contact angles using DI H<sub>2</sub>O and DMSO were measured on release film, an inert and smooth surface
- Results show no time dependence and thus viscosity effects are dismissed as a possibility for this affect
- Average Contact Angle Measurement 100 80 (degrees) 0 seconds 60 5 seconds 10 seconds 40 30 seconds 60 seconds 20 0 DI H2O DMSO Fluid
- Possibly due to adsorption of fluid onto surface

### JMS Peel Ply Texture Orientation





 Effect of surface texture left in epoxy resin after peel ply removal



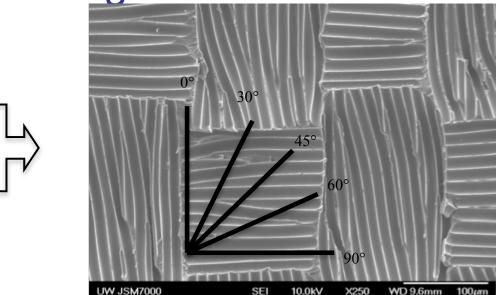
### JMS Peel Ply Texture Orientation

Camera





 Peel ply angle is defined as the angle at with the peel ply texture is oriented with respect to the goniometer camera

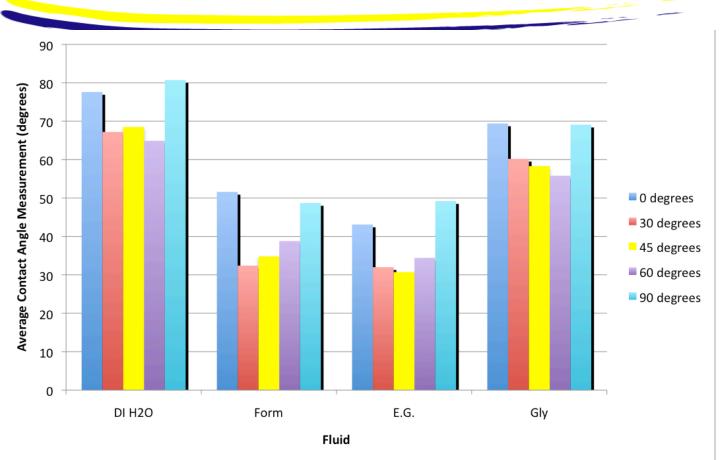


#### Effect of Peel Ply Texture Orientation on Contact Angle

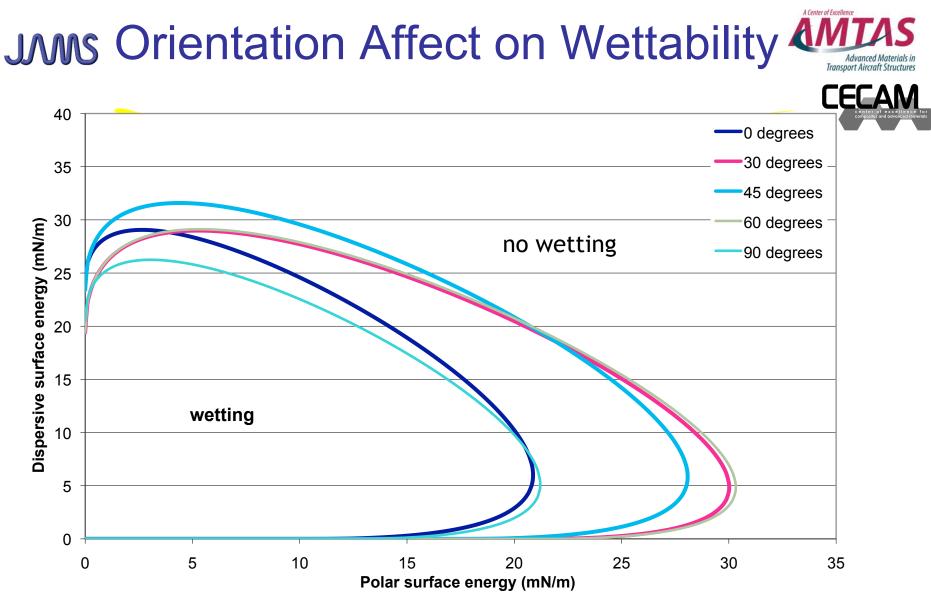
JMS



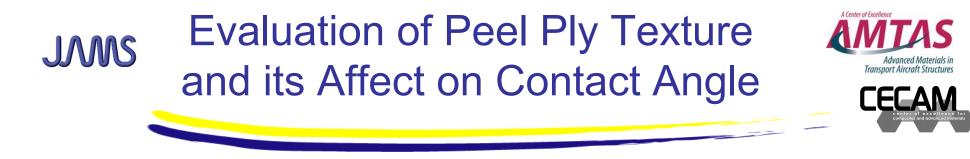




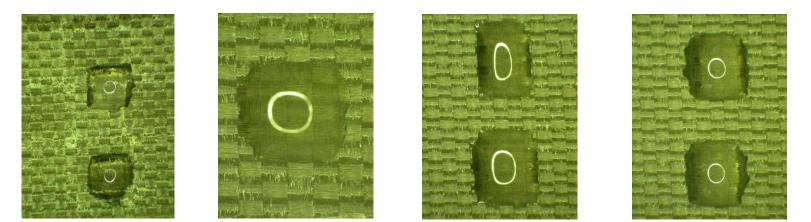
- Peel ply orientation affects contact angle measurement
  - Contact angles measured at 0 and 90 degrees are greatest



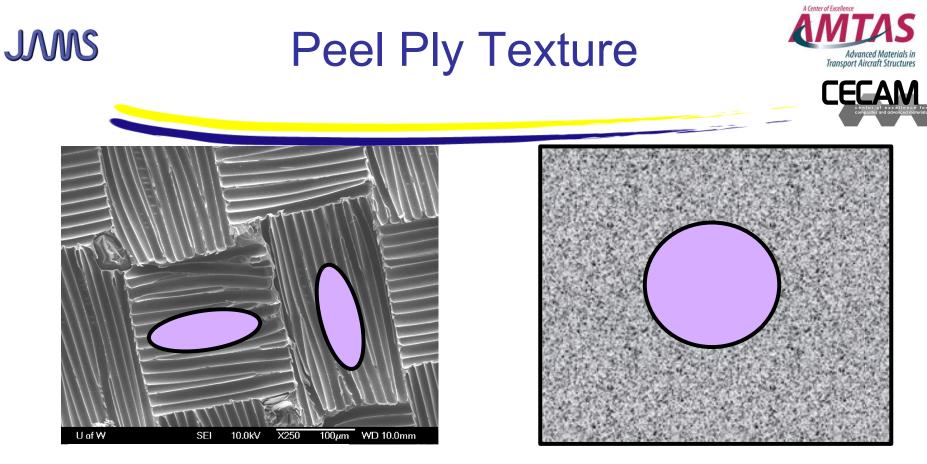
30 and 60 Orientation produced similar wettability envelopes 0 and 90 Orientation produced similar wettability envelopes



- Difference in contact angle at differing orientations of the substrate is due to the texture left in the resin upon peel ply removal
  - The fluids form non-circular drops on the substrate



Top down pictures from right to left of formamide, DI water, ethylene glycol, and glycerol drops (note: white circles on each drop are a reflection of light)

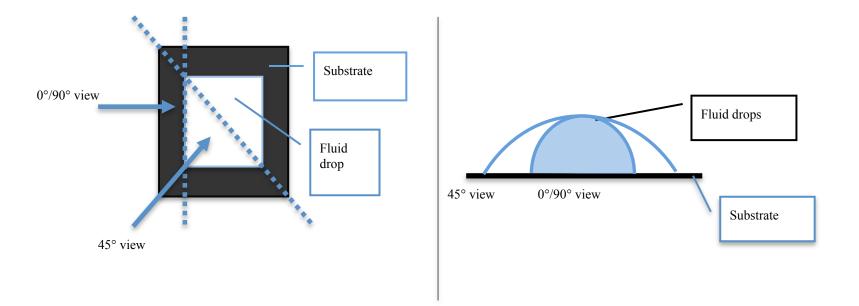




 Peel ply texture affects the shape of the fluid drop



• The non-circular drops resulted in differing contact angle measurements at different orientations





## How sensitive is the CA method to contamination?

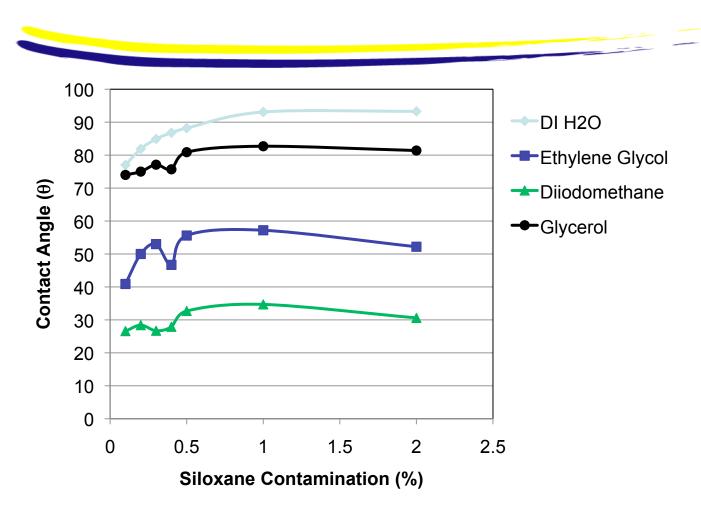
- Controlled amount of siloxane added to peel plies to produce intentionally contaminated CFRP surfaces for evaluation
  - Contact Angle (4 fluids)
  - Surface Energy and Wettability
  - Bond Quality (G<sub>IC</sub> and fracture mode)



### Contact Angle vs. Si



CECAN



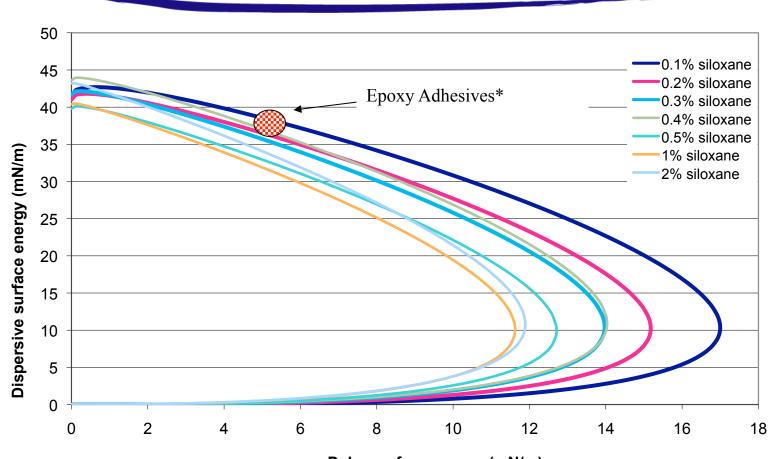
Contact Angle Increases with Increasing Si



### Wettability Envelopes

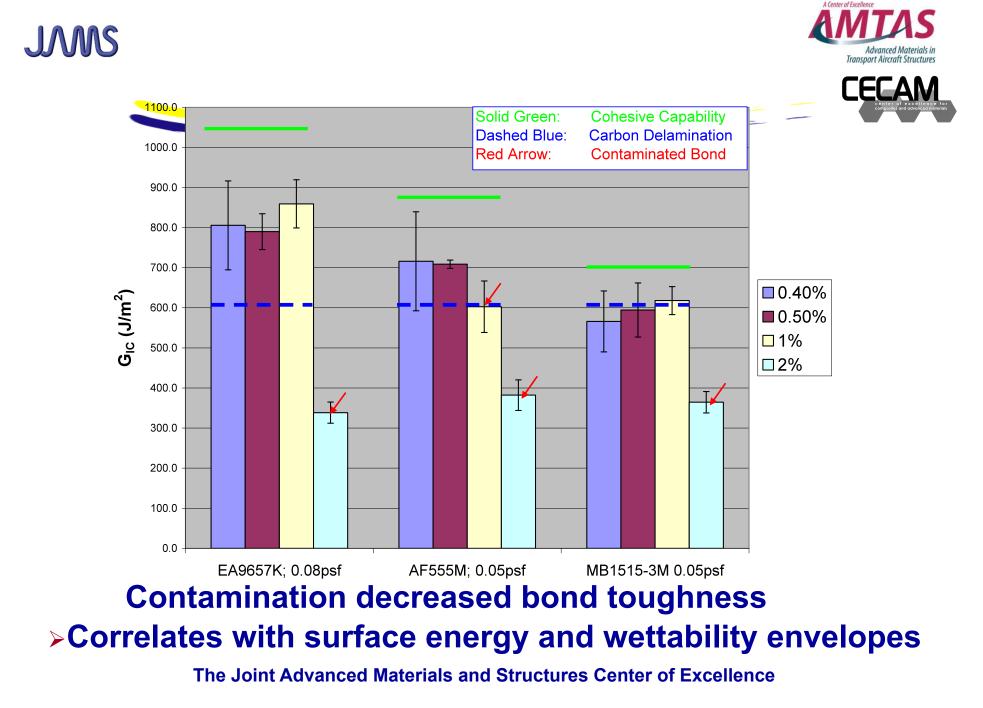






Polar surface energy (mN/m)

Wettability envelopes decreased with increasing Si





- Investigation of Other Experimental Variables
- Modeling of Texture affect
- Advancing/Receding Contact Angle Measurement
- Other Surface Energy Measurement Techniques
- Effect of CA fluids on bonding
- Other Contaminates? (input requested)
- Identification of Preferred Fluids



- Time to measure contact angle has an affect on the measurement
  - Some fluids more sensitive
- Surface texture can results in non spherical drops
  - results in change in the measured contact angle
- CA, wettability and bond quality correlated with % Si
- Use of multiple fluids and wettability envelopes recommended
- Potential QA technique for surface preparation



### A Look Forward





- Benefit to Aviation
  - Better understanding of peel ply surface prep.
  - Guide development of QA methods for surface prep.
  - Greater confidence in adhesive bonds
- Future needs
  - Surface energy (wetting) vs. bond quality model
  - Surface energy at cure temperature
  - QA method to ensure proper surface for bonding
  - Applicability to other composite and adhesive systems
  - Model to guide bonding based on characterization, surface prep. and material properties



### Acknowledgements





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- Advanced Materials in Transport Aircraft Structures

A Center of Excellence

- Boeing Company
- Precision Fabric Grc ...
- Richmond Fabrics
- Airtech International
- Prof. Mark Tuttle (UW)



BOEING



- Crucial for proper adhesion in composites
- Several methods
  - Peel ply (as tooled)
  - Abrasion (Sanding or grit blasting)
- Surface preparation influences surface energy and the wettability of a surface, also prevents/removes contamination
- A high energy surface promotes intimate contact between the surface and the adhesive





### QUESTIONS ? COMMENTS? SUGGESTIONS?