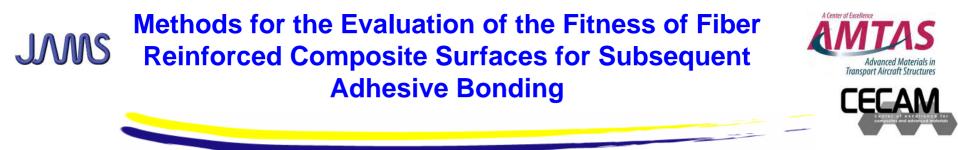


Methods for the Evaluation of the Fitness of Fiber Reinforced Composite Surfaces for Subsequent Adhesive Bonding Bill Stevenson Dept. of Chemistry Wichita State University









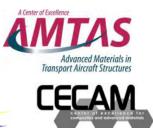
- Motivation and Key Issues
- With user safety in mind, the FAA has determined to coordinate and lead an effort to improve our understanding of current composite bonding methodology in the commercial aerospace and scientific community, with special emphasis on surface preparation and characterization prior to the bonding event.
- Objectives
- To develop nondestructive analytical techniques for application to the analysis of prebonded CF reinforced composite surfaces on the shop floor that can be used to support a decision to "go ahead" and bond the part or "stop" and have it re-cleaned, dried, or cured further prior to bonding
- To integrate these techniques of analysis into the joint FAA/NIAR scarf repair project to develop insights into the influence of surface contaminants on the integrity of composite patches for very thick CF reinforced composite laminates to be used in primary load bearing applications in next generation aircraft
- Approach Techniques of analysis
- Contact angle measurement Overall fitness for bonding
- Near Infrared Spectroscopy Water
- Laser desorption mass spectroscopy Specific surface contaminents such as jet fuel, hydraulic oil, etc.



FAA Sponsored Project Information



- Principle Investigators & Researchers
 - William Stevenson, Irish Alkalen, Laura Stevenson, Michael VanStipdonk, Dennis Burns (Chemistry, Wichita State University) Chemical analysis
 - Lamia Salah, John Tomblin (National Institute for Aviation Research – NIAR, Wichita State University) Sample design, fabrication, mechanical testing, design of and interpretation
 - Charles Yang (NIAR & Aeronautical Engineering, Wichita State University) Finite element analysis
- FAA Technical Monitor
- Curt Davies
- Other FAA Personnel Involved
 - Larry Ilcewicz
- Industry Advisory
- Michael Borgman, Robert Bohaty



Water by near IR Spectroscopy

- Dissolved entrained water in the composite will migrate to the surface and produce a poor bond.
- Even if the surface is dried by transient heating, water can migrate from the bulk to the surface to cause a poor bond
- Infrared spectroscopy can detect water if signals from water and hydroxyl groups in the resin can be separated – near infrared spectroscopy
- Instrument makers will custom build near IR configurations for the shop floor



•Phase I (Dec 2005) goal

•To have a working Integrating Sphere diffuse reflectance Near IR system in place to analyze surface treated composite samples for water content

Progress: Completed

 Integrating sphere module was custom built to fit Nexus spectrometer.

Hardware and software "glitches" were solved.
Preliminary testing indicated that water in high performance composites containing Toray T-800 fibers can be observed by this method



Phase II (2006) goals

 To produce calibration curves relating spectrum to water content in epoxy resin based composite systems

Progress: Completed

- Calibration curves have been constructed for...
- A medium performance epoxy resin
- A high performance epoxy resin
- A carbon fiber reinforced composite containing the same high performance epoxy resin



To integrate this technique into the joint FAA/NIAR scarf repair project to characterize "conditioned" repair surfaces made from high performance composite prior to repair layup for water content at and near the scarfed surface

Progress: To commence this summer



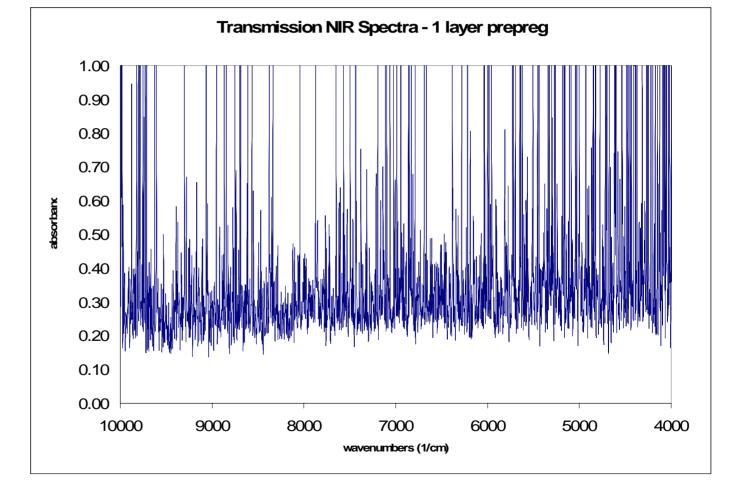
Background Info

Combination stretching/deformation band @ 5215cm⁻¹ $v_{assym} + \delta_{in plane}$ Pro's

Free of interference from resin hydroxyls Quantitative measure of water Con's Little info on state of water in resin

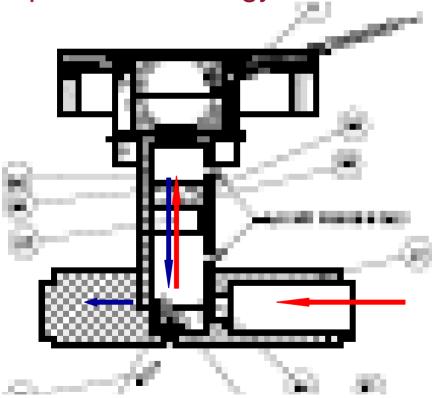


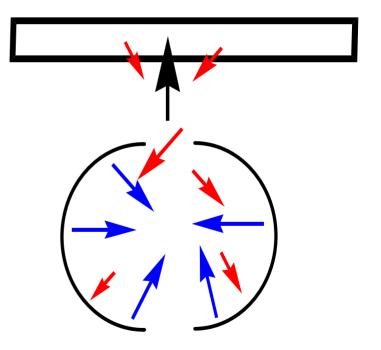
Unfortunately near IR spectroscopy of composites does not work in transmission





Diffuse reflectance Near IR spectroscopy using integrating sphere technology





TE Nexus with Integrating sphere module



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A composite sample placed on top of the integrating sphere module and ready for analysis

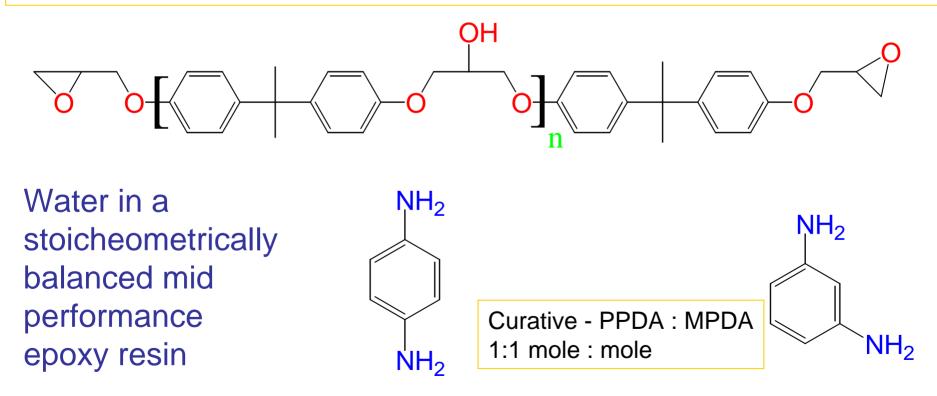




- Procedure for resin samples
- (1). Resin cured between steel plates
- (2). Near IR spectrum acquired
- (3). Weighed then exposed to water vapor @ 140 °F
- (4). Reweighed then near IR spectrum acquired
- (5). Dried @ 50°C, reweighed, and near IR spectrum acquired

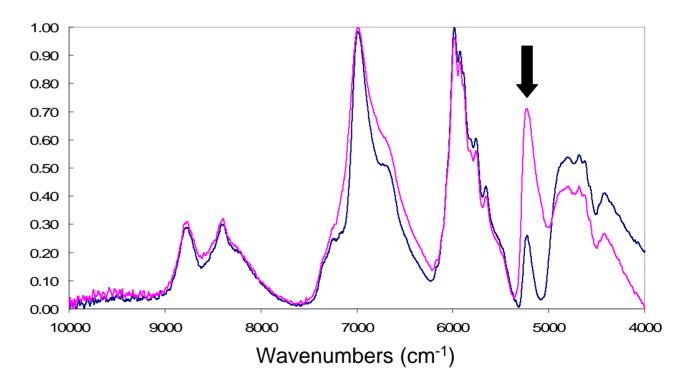


Cured 2h at 120°C then 2h at 180°C and stored in a dessicator



NIR (integrating sphere) Spectra - resin825-MPDA-PPDA dry and wet

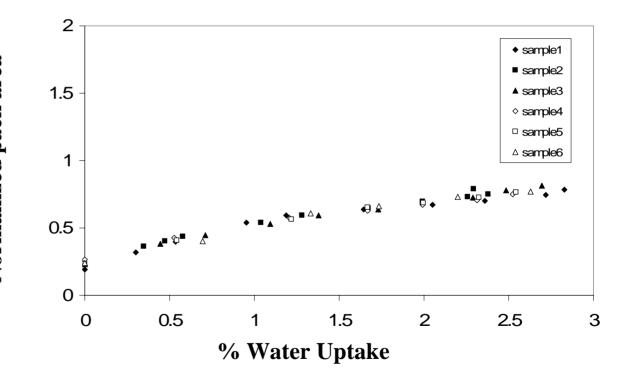
Near IR spectra of mid performance DGEBA epoxy resin before (blue) and after (wine) water absorption. The water absorption (overtone) is denoted by the arrow.



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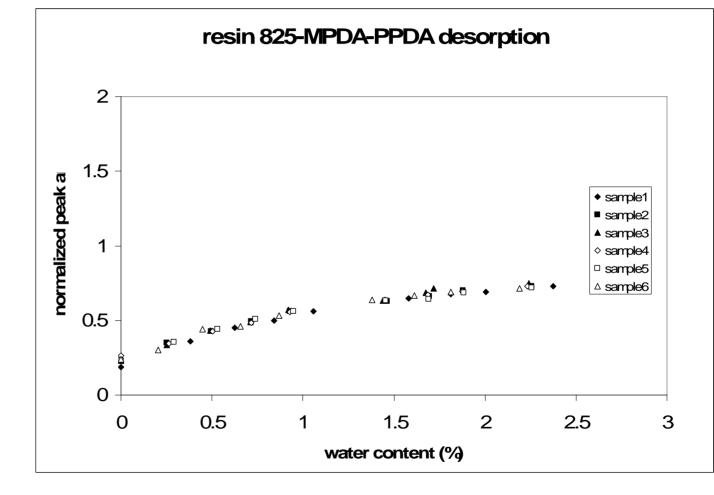
resin825-PPDA-MPDA water uptake

A plot of normalized water absorbance at 5215 cm⁻¹, versus % water uptake for this resin system

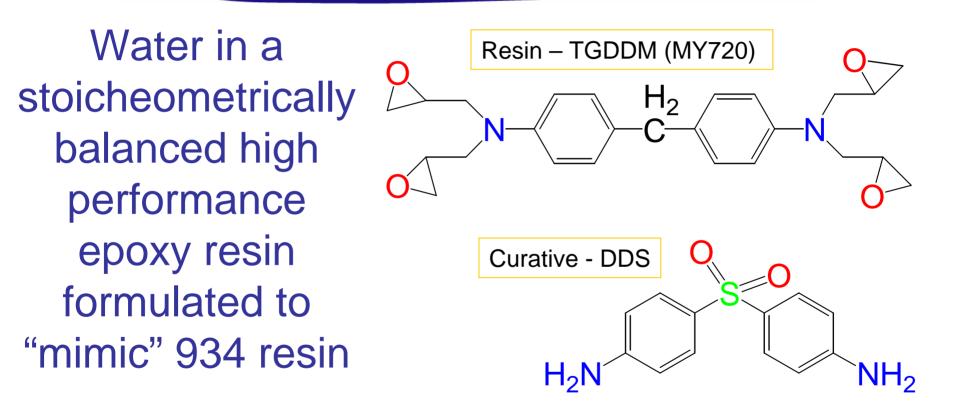


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Spectra were also obtained during desorption (drying) cycles @ 50°C



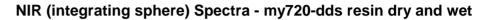
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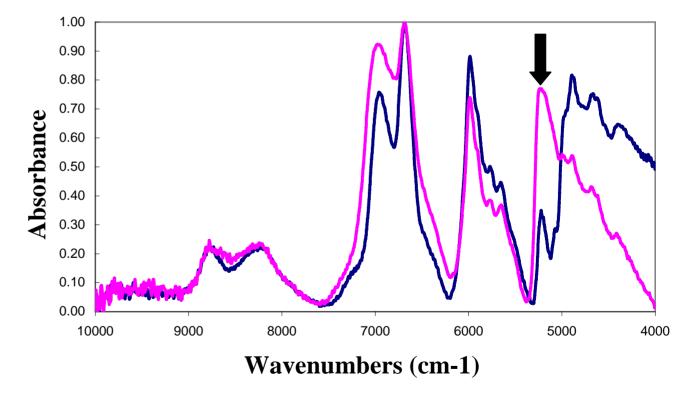
Resin cured between steel plates 2h at 140°C then 4h at 200°C

- HAN





Near IR spectra of a high performance epoxy resin containing MY720 epoxy resin and DDS curative before (blue) and after (wine) water absorption. The water absorption (overtone) is denoted by the arrow



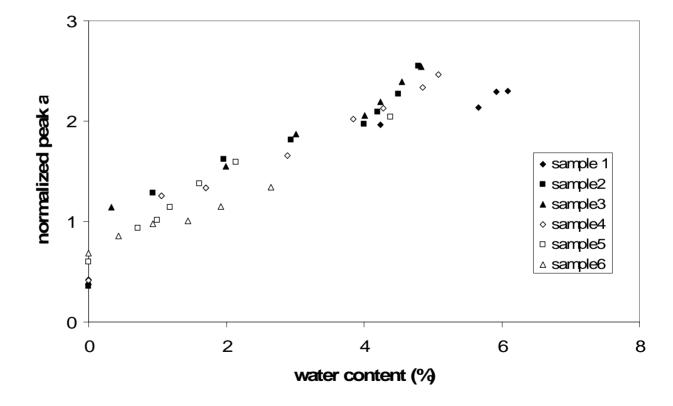
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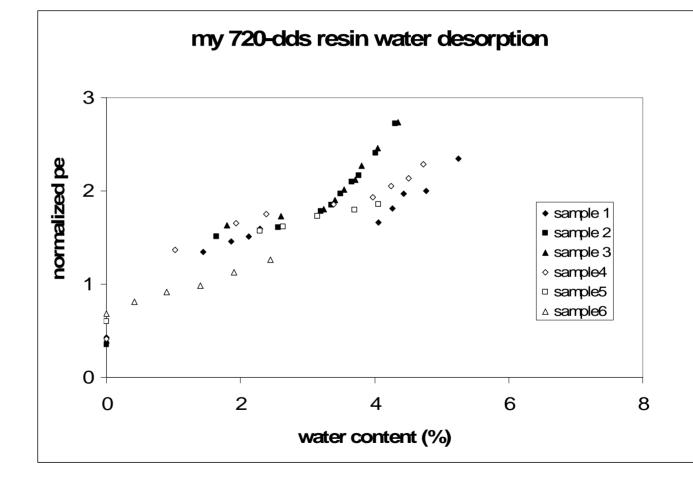


my720-dds resin water uptake

A plot of normalized water absorbance at 5215 cm-1 versus % water uptake for the MY720-DDS resin system



Spectra were also obtained during desorption (drying) cycles @ 50°C



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Conclusions

- More water is absorbed by the high performance resin than the mid performance resin (free volume)
 More scatter with the high performance regin then with the mid performance
- resin than with the mid performance resin
 Reproducible calibration curves



Water in an epoxy resin – carbon fiber based composite

Fibers – woven T300 carbon

Resin – generic "934" high performance resin

Details of cure – double vacuum bagged and cured in a forced air oven, 2h at

120°C followed by 2h at 200°C

Component	Percent Composition by Mass	Methods
Carbon Fiber	62.0 - 63.1%	solvent extraction & HNO ₃ resin burn off
TGDDM (MY720 epoxy resin)	22%	¹ H NMR spectroscopy, prep TLC
DDS (amine curative)	14%	¹ H NMR spectroscopy, prep TLC
BF ₃ (catalyst)	< 1 %	¹¹ B & ¹⁹ F NMR spectroscopy
Acetone solvent	< 1%	mid IR & ¹ H NMR spectroscopy

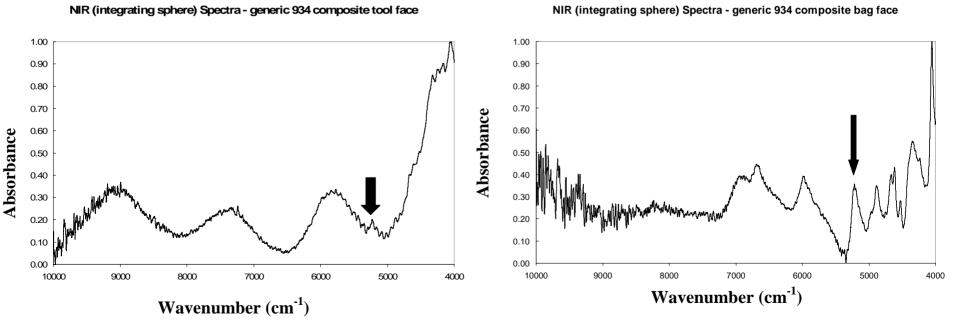


Strength Works well with fiber reinforced composite on resin rich vacuum bag and peel ply surfaces

Weakness

Does not work well on resin poor (tool) surfaces

Near IR spectra of tool face (left) and bag face (right) of composite. "Bag face" spectra can be normalized

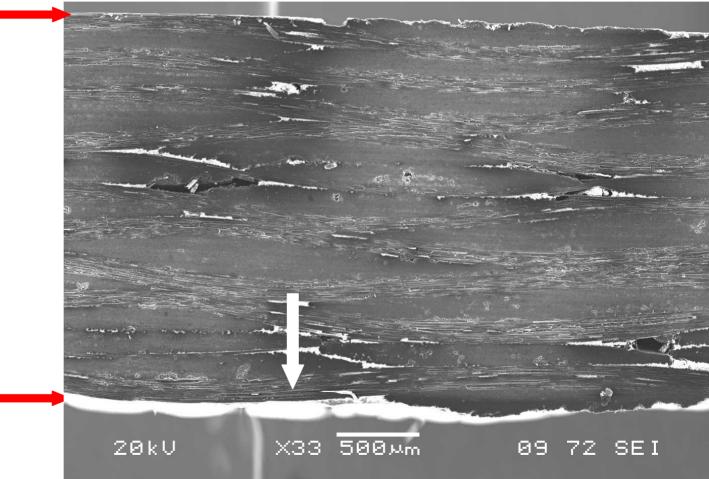


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Tool face

Resin rich regions are overexposed due to charging

Bag face



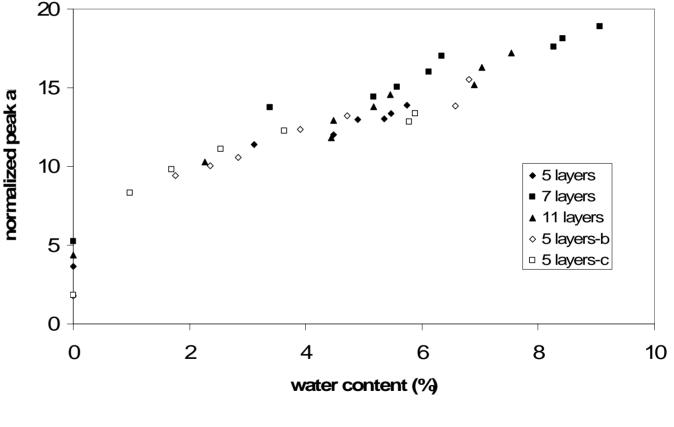
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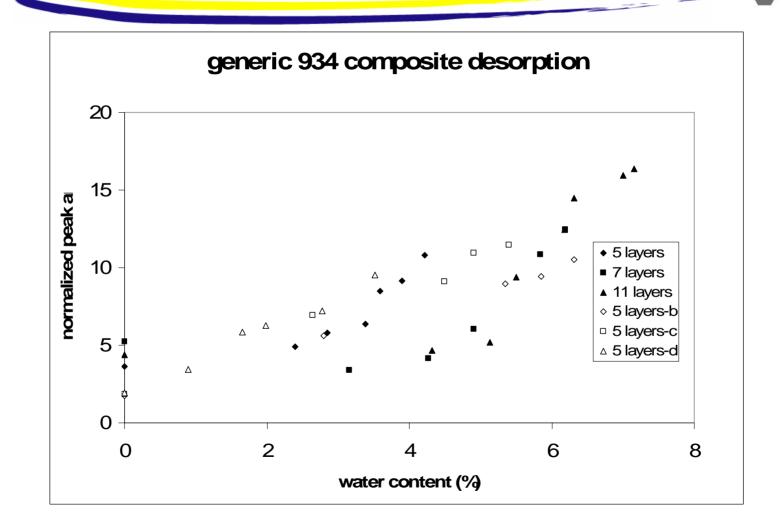
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generic 934 composite water uptake

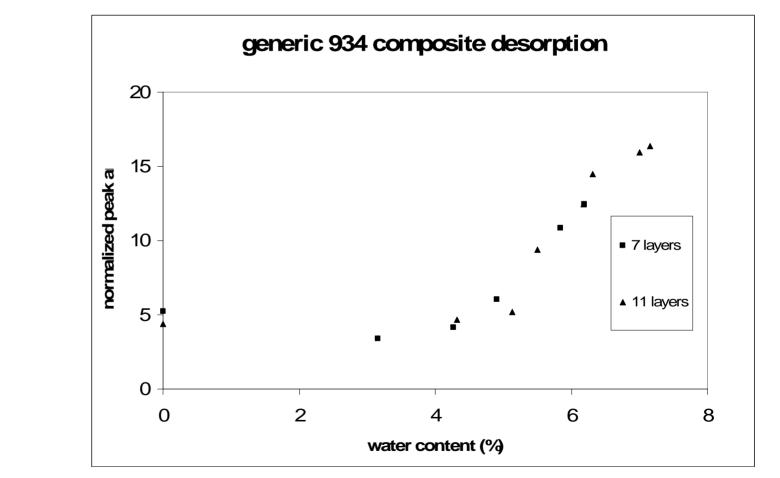
A plot of normalized water absorbance at 5215 cm⁻¹ versus % water uptake measured at the bag face of a composite made using generic "934" resin and carbon fiber fabric





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Methods for the Evaluation of the Fitness A Center of Excellence JMS of Fiber Reinforced Composite Surfaces for Subsequent Adhesive Bonding generic 934 composite desorption 20 15 normalized peak a 5 5 layers 10 layers ◊ 5 lavers-b □ 5 layers-c Δ Δ 5 △ 5 layers-d Δ 0 0 1 2 3 4 5 6 7 water content (%)



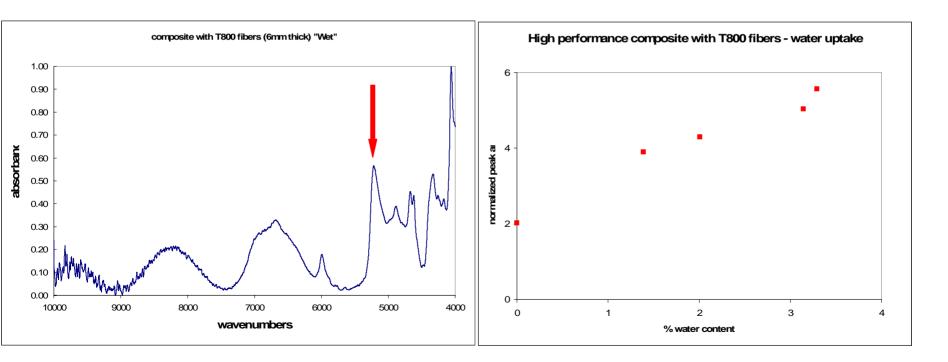
7 & 11 layers

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Conclusions

Scatter during absorption of water by the composite < scatter during water loss
Hysteresis during fast water loss due to concentration gradient imposed by fiber layers.
At similar overall water content measured during a fast desorption cycle, the thicker composites contain less water at the surface and produce a smaller absorption

High performance composite with Toray T-800 fibers



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Resin and adhesive cure state by diffuse reflectance (integrating sphere) near IR spectroscopy Preliminary results presented here

Adhesive water uptake studies In progress

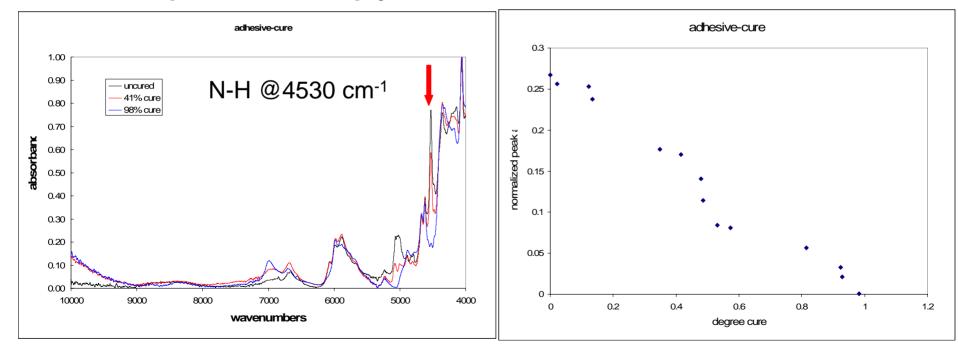


Composition of a filled high shear adhesive

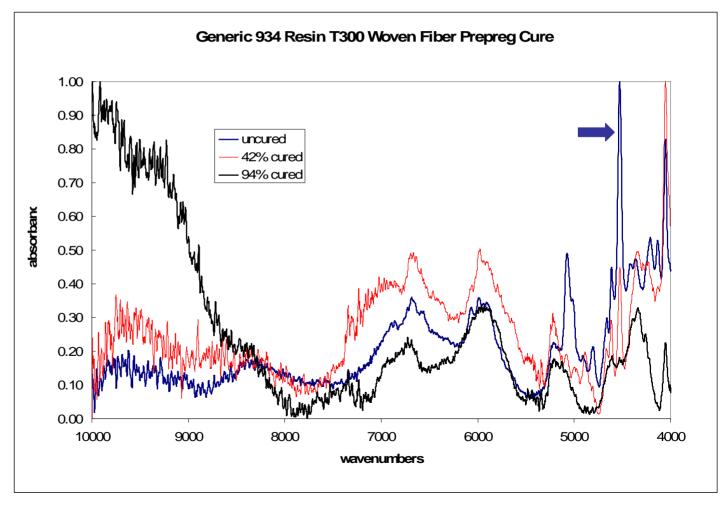
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Filled High Shear Adhesive cure state by near IR spectroscopy



934 resin *w* T300 woven fibers prepreg cure state

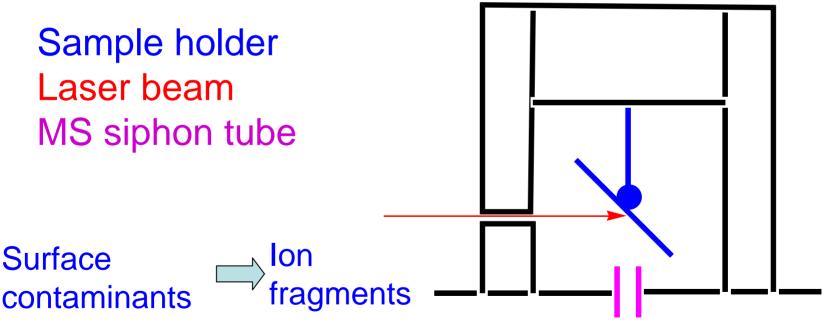


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Phase I (Dec 2005) goals

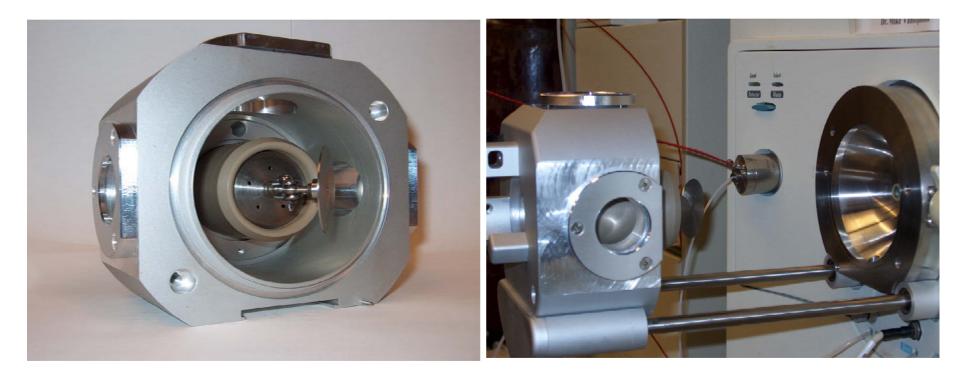
- To have a working LD-MS system in place to analyze surface treated composite samples
- Progress
- Completed laser has been purchased, sample stage has been built and "debugged". System works.
- **Commenced work on Phase II**
- **Development of spectral database for analysis of unknowns**



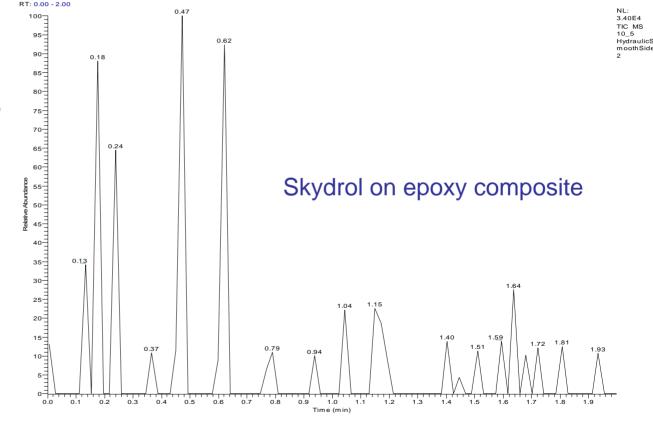
To integrate this technique into the joint FAA/NIAR scarf repair project to characterize "conditioned" scarfed repair surfaces made from high performance composite prior to repair layup for hydraulic fluid and jet fuel at the scarfed surface

Progress: To commence this summer





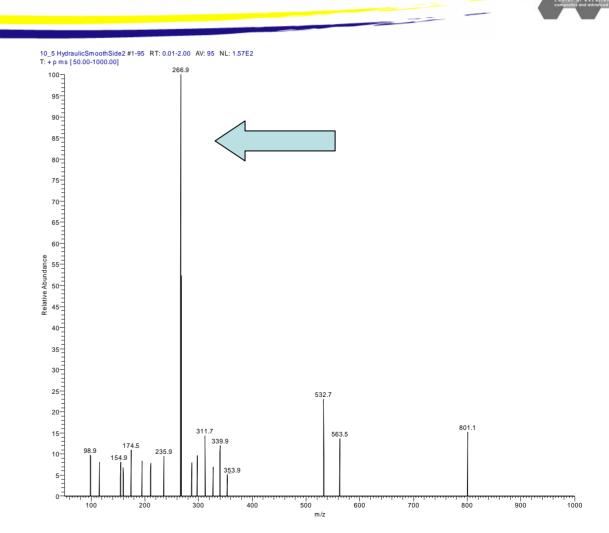
Material is removed from the surface in "bursts" We can focus on any burst or integrate the mass spectrum over the entire time interval. Molecular damage is minimal & parent ions can be identified



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Transport Aircraft Structure

Time averaged spectrum with tributyl phosphate (TBP) M266 arrowed



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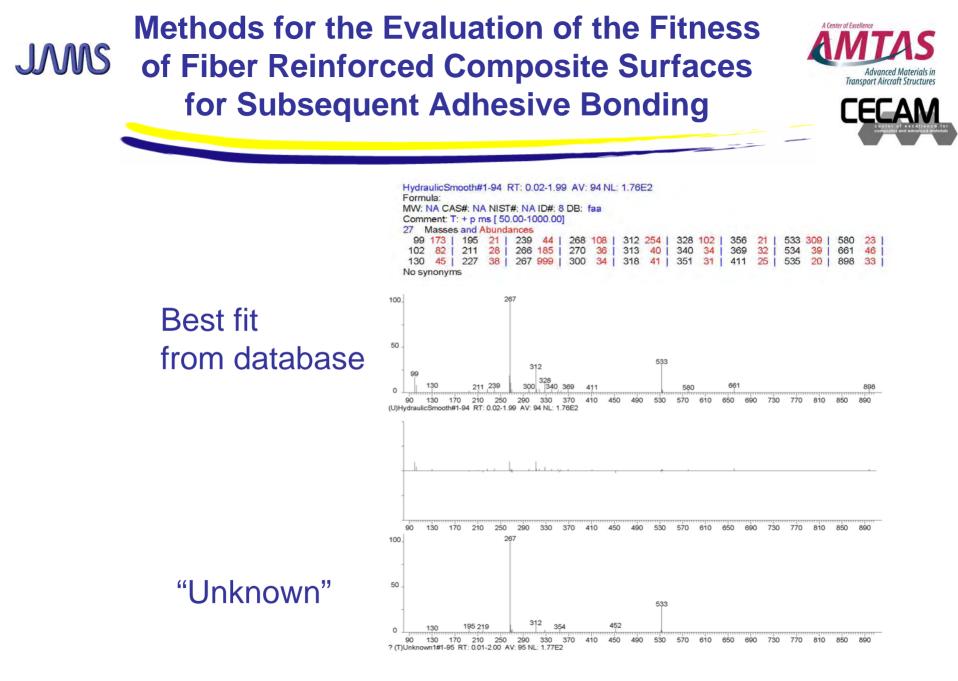
Transport Aircraft Structure



Spectra can be integrated into the NIST library using NIST MS search program (Version 1.7) tool to produce a library of LD-MS spectra to compare with and help recognize unknowns

Proof of concept

- •Acquire spectrum of skydrol on composite
- •Add to NIST library
- •Reacquire as "unknown" on composite, search library for best fit, and re-identify Skydrol as most probable species



- 1 611 663R 98.0P (U) HydraulicSmooth#1-94 RT: 0.02-1.99 AV: 94 NL: 1.76E2
- 2 306 422R 0.98P (U) hexa-ethyleneglycol#1-446 RT: 0.01-3.99 AV: 446 NL: 4.00E2
- 3 286 315R 0.44P (U) Sulfisoxazole
- 4 273 278R 0.28P (U) Adenosine
- 5 240 302R 0.07P (U) 3-Methylcholanthrene
- 6 232 307R 0.05P (U) Methanone, bis[4-(dimethylamino)phenyl]-
- 7 224 277R 0.04P (U) Ergotamine Tartrate
- 8 214 251R 0.02P (U) Benzenamine, 4,4'-methylenebis[2-chloro-
- 9 204 232R 0.02P (U) Octadecanoic acid, methyl ester
- 10 188 192R 0.01P (U) Propyl Gallate
- 11 152 178R 0.00P (U) Squalane
- 12 137 156R 0.00P (U) Stannane, (acetyloxy)triphenyl-

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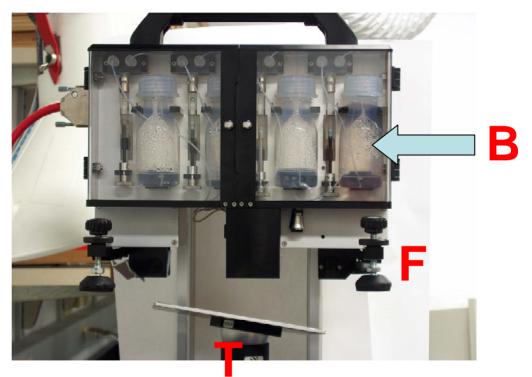
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Surface free energy by contact angle measurement H carry handle

- **F** feet
- B solvent bottle
- T tilting stage







Phase I (Dec 2005) goals

To have a working automated portable multi-solvent contact angle measuring device in place to analyze surface treated composite samples as supplied by NIAR and Spirit Aerosystems beginning

Progress

Completed- Device has been fabricated and delivered (FDS PCA 100M-4). Sample mount has been modified to handle wedge shaped surfaces. Contact angle measurements have been made on scarfed surfaces ground into sheets of high performance composite.



To integrate this technique into the joint FAA/NIAR scarf repair project to characterize "conditioned" scarfed repair surfaces made from high performance composite prior to repair layup for surface free energy at the scarfed surface

Progress: To commence this summer



Selected results

SURFACE	FLUID	CONTACT ANGLE (water)	SURFACE ENERGY - EOS (mN/m)
rough	none	59.2	48.28
smooth	none	65.3	44.57
wedge	none	47.5	55.16 <
smooth	deicing fluid	49.5	54.05
wedge	deicing fluid	45.5	56.29
rough	skydrol (hydraulic fluid)	47.2	55.33
smooth	skydrol (hydraulic fluid)	60	47.80
wedge	skydrol (hydraulic fluid)	66.5	43.83 <

rough	JP-8	37.5	60.66
smooth	JP-8	53.4	51.74
wedge	JP-8	45.3	56.41 <
	saltwater	no measurable contact angle, liquid simply spreads on contact	



LD-MS: An analytical tool to provide surface analysis at atmospheric pressure of composite and metal surfaces contaminated with volatile organics

Contact angle device: A robust, portable, user friendly version of a device that can be mounted on a cart or robotic arm to contribute to a go/no go answer to the bonding question

Near IR diffuse reflectance: A device that can be mounted on a hand held grip to determine moisture content and surface cure state in a composite part

NIST repair project : Design of repair methodologies for thick composite parts in the field